

Accelerated acoustic prediction of aging and failure

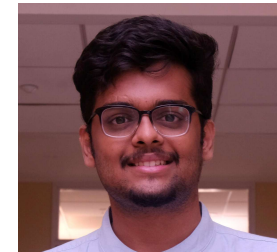


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Wet Systems (Project Review #3, 2026)

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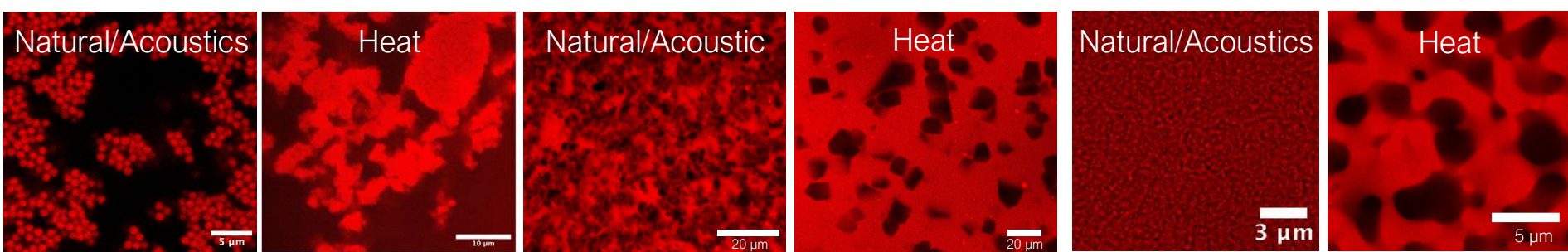
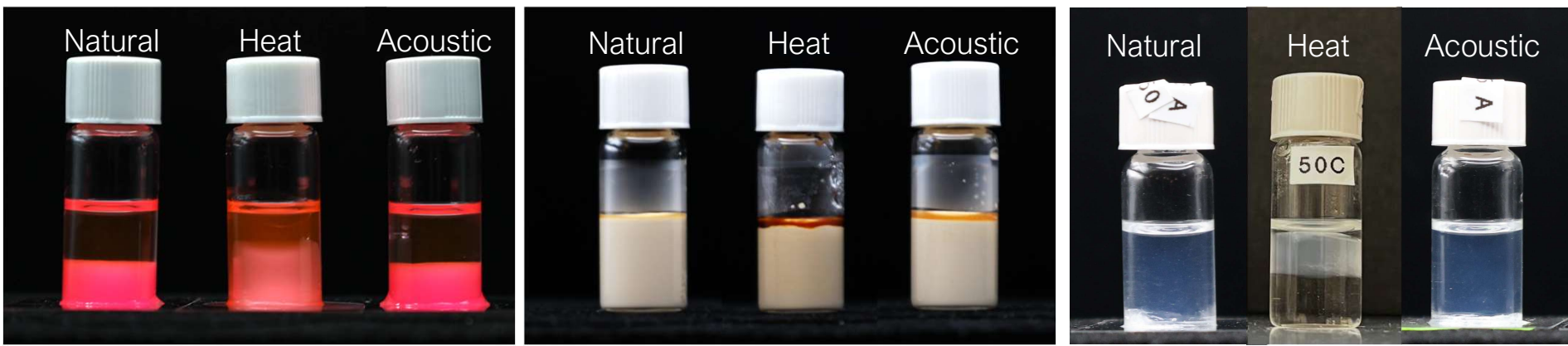
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Acoustics accelerate natural aging of wet systems

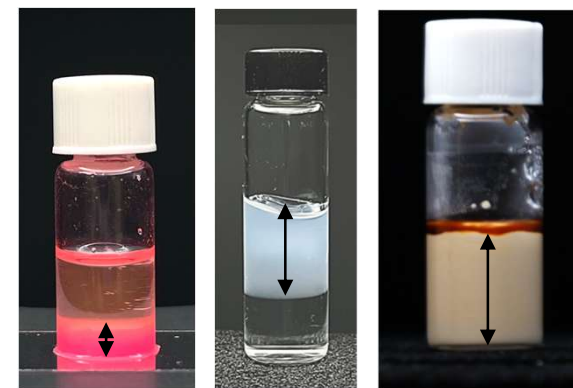
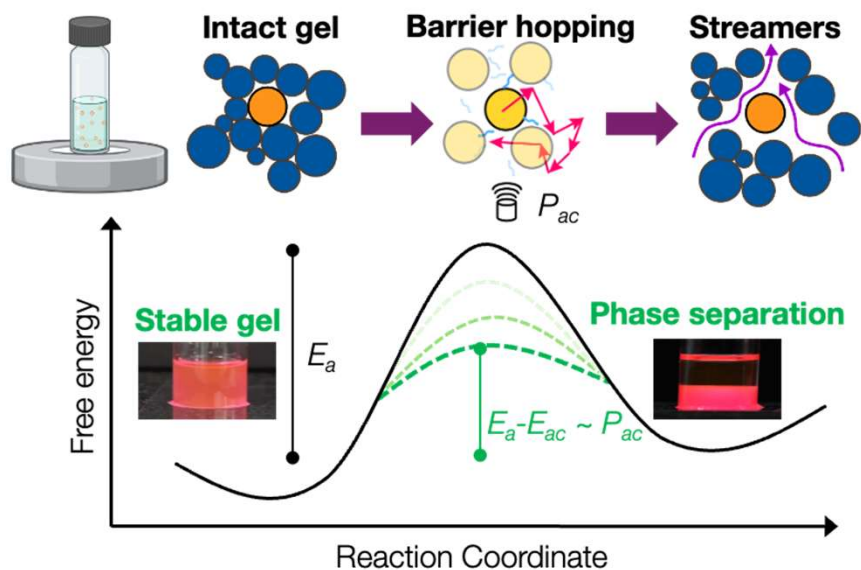
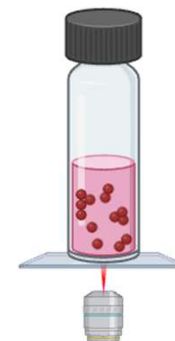
Colloidal particle dispersions Commercial suspension concentrates Oil/water nanoemulsions



Acoustics as an alternative aging method

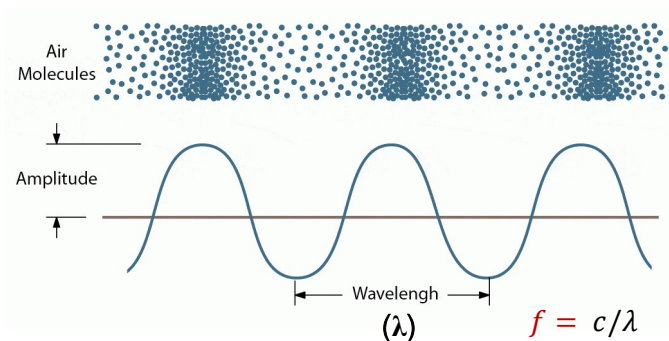
Mechanism: Acceleration dominated by acoustic streaming effects

- Acoustic power range: 0.1 - 300 mW (high end)
- Acoustic pressure range: 1 - 30 kPa
- Minimal attenuation from viscous or interfacial losses
- Within the 0.1 mW range, phase separation acceleration factors $\sim 5x$



Track microstructure and phase separation

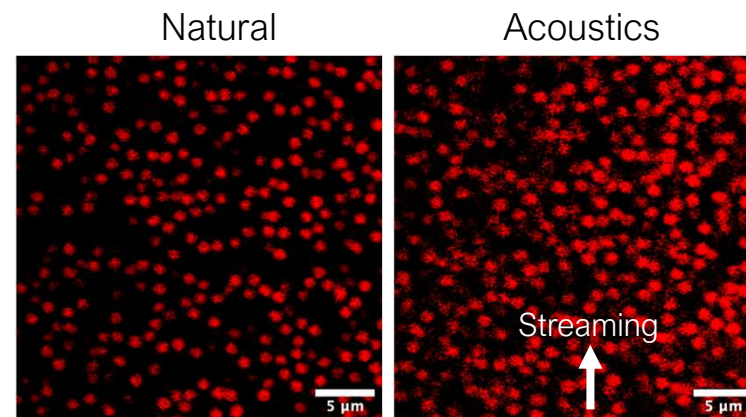
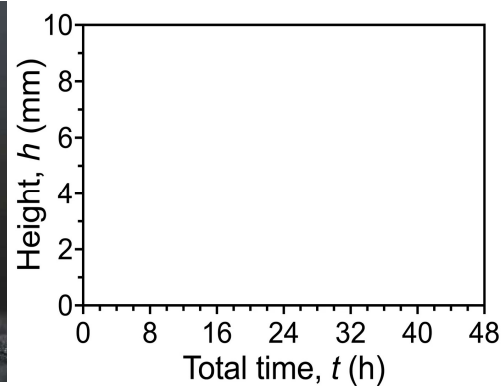
Acoustics as an alternative aging method



Acoustic power range: 0.1 - 300 mW

Acoustic pressure range: 1 - 30 kPa

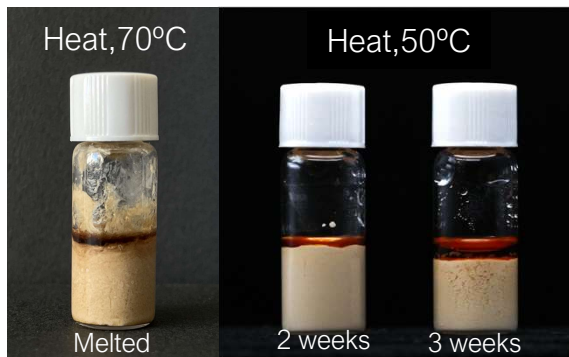
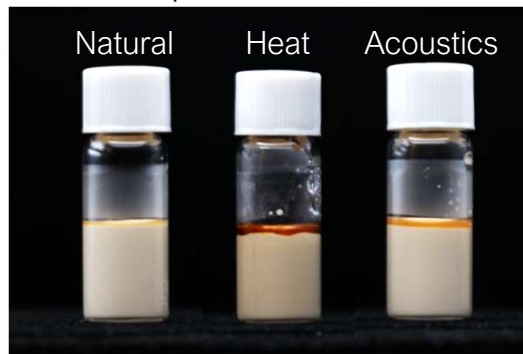
	PMMA gels	Nanoemulsions
Attenuation/scattering/cavitation	-	-
Acoustic transmittance	0.7	0.8
Radiation pressure potential	$0.05 kT$	$10^{-6} kT$
RP velocity ($\mu\text{m/s}$)	10^{-7}	10^{-10}
Streaming velocity ($\mu\text{m/s}$)	2.0	1.3
Peclet number (particle)	5.6	0.01
Mechanism	Streaming	Brownian/interfacial



Acoustic vs. heat treatment of commercial suspensions

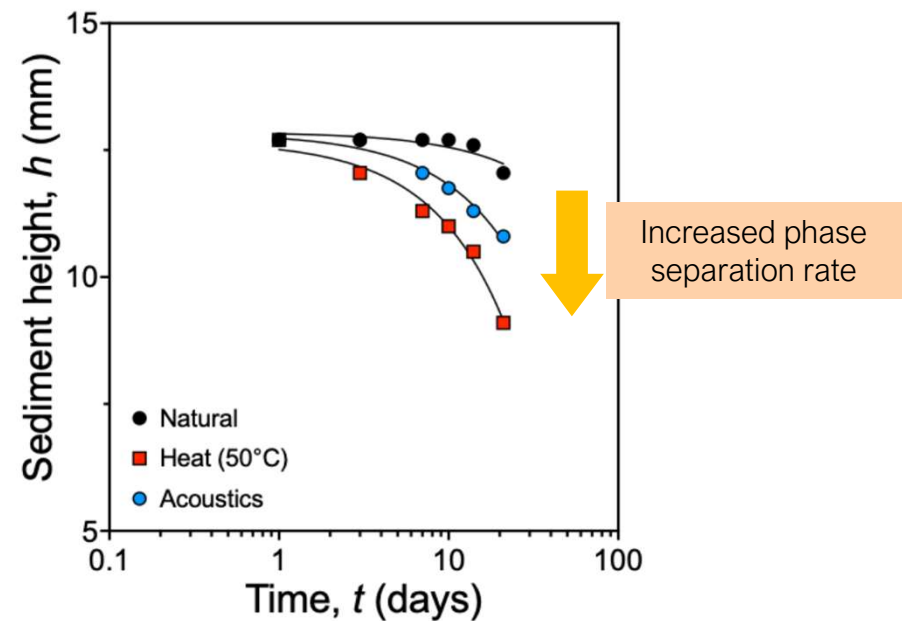
Heating is fast, but causes unnatural changes in formulations

IFPRI suspension concentrates



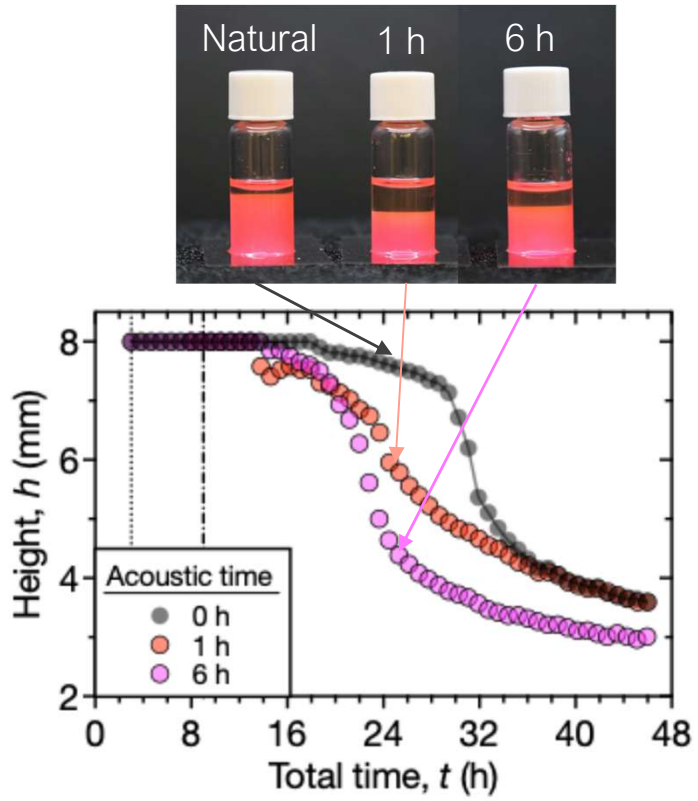
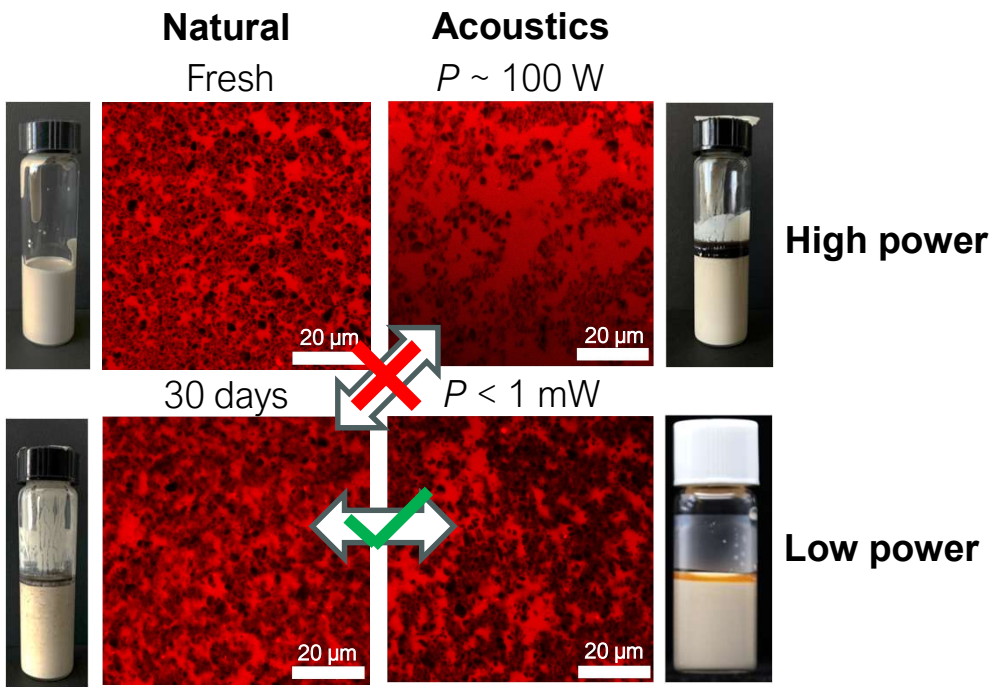
Degradation temp = 54°C

Acceleration factor from heating: 10-15x
Acceleration factor from acoustics: 5-6x

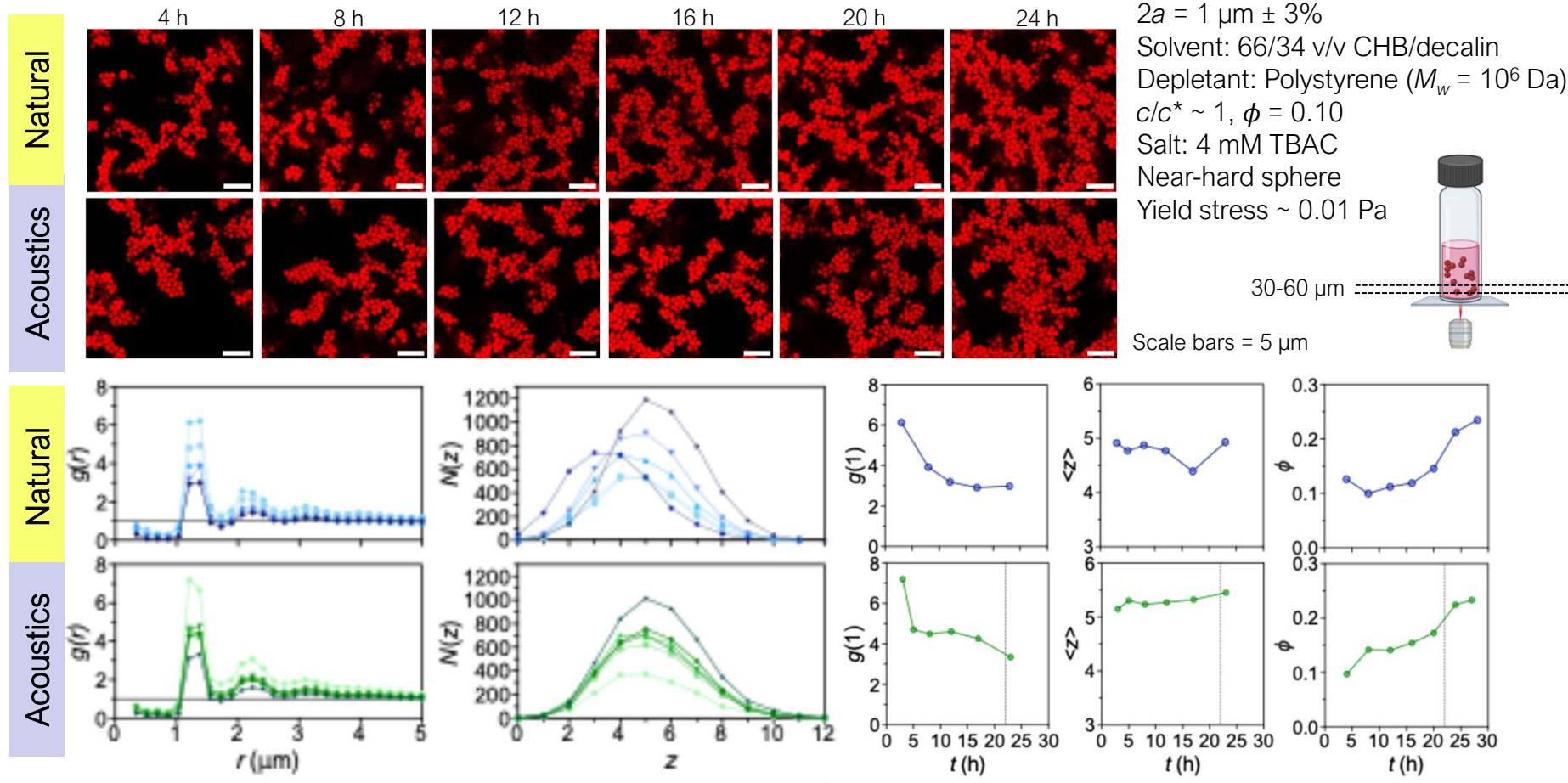


Why not simply increase the acoustic exposure?

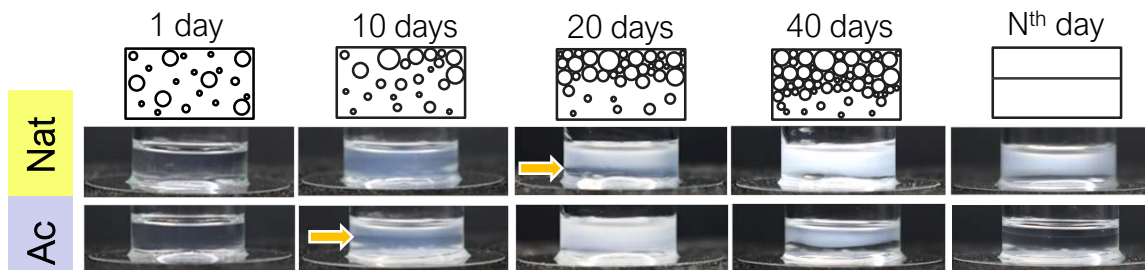
- Possible, but overly high power leads to local heating and deterioration of samples
- Nonlinear effect (i.e. more power/exposure does not mean faster)
- This is seen in both PMMA gels and oil/water nanoemulsions



Structural analysis on colloidal gels



Effect of acoustic power and frequency on emulsions



PDMS oil/water nanoemulsions

$2a \sim 40 \text{ nm} \pm 30\%$

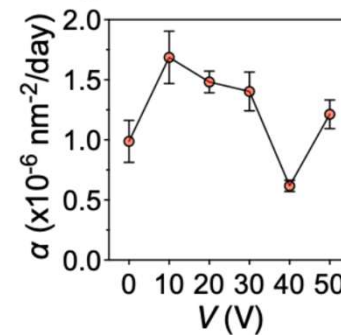
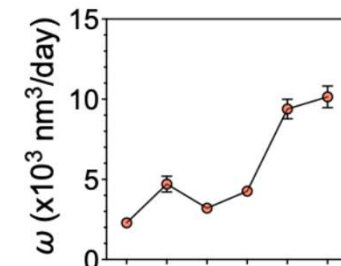
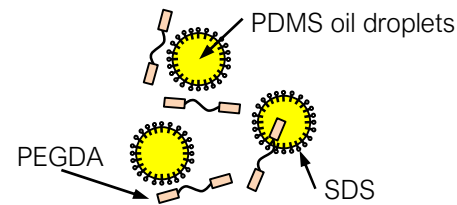
Dispersed phase: PDMS (5 cSt), $\phi = 0.20$

Continuous phase: De-ionized water

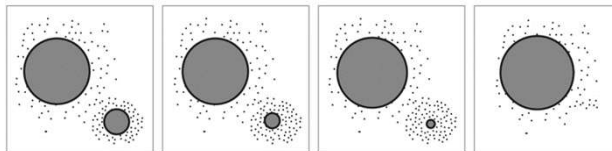
Surfactant: 200 mM sodium dodecyl sulfate

Co-surfactant: 33 vol% PEGDA ($M_w = 700\text{Da}$)

Yield stress $\sim 1 \text{ Pa}$



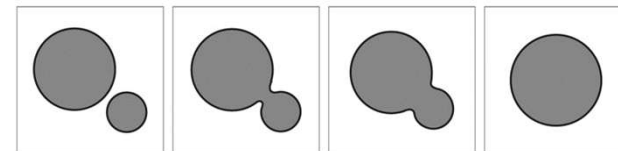
Ostwald ripening



$$\langle r_H \rangle \sim t^{\frac{1}{3}}$$

Self-similar PSD

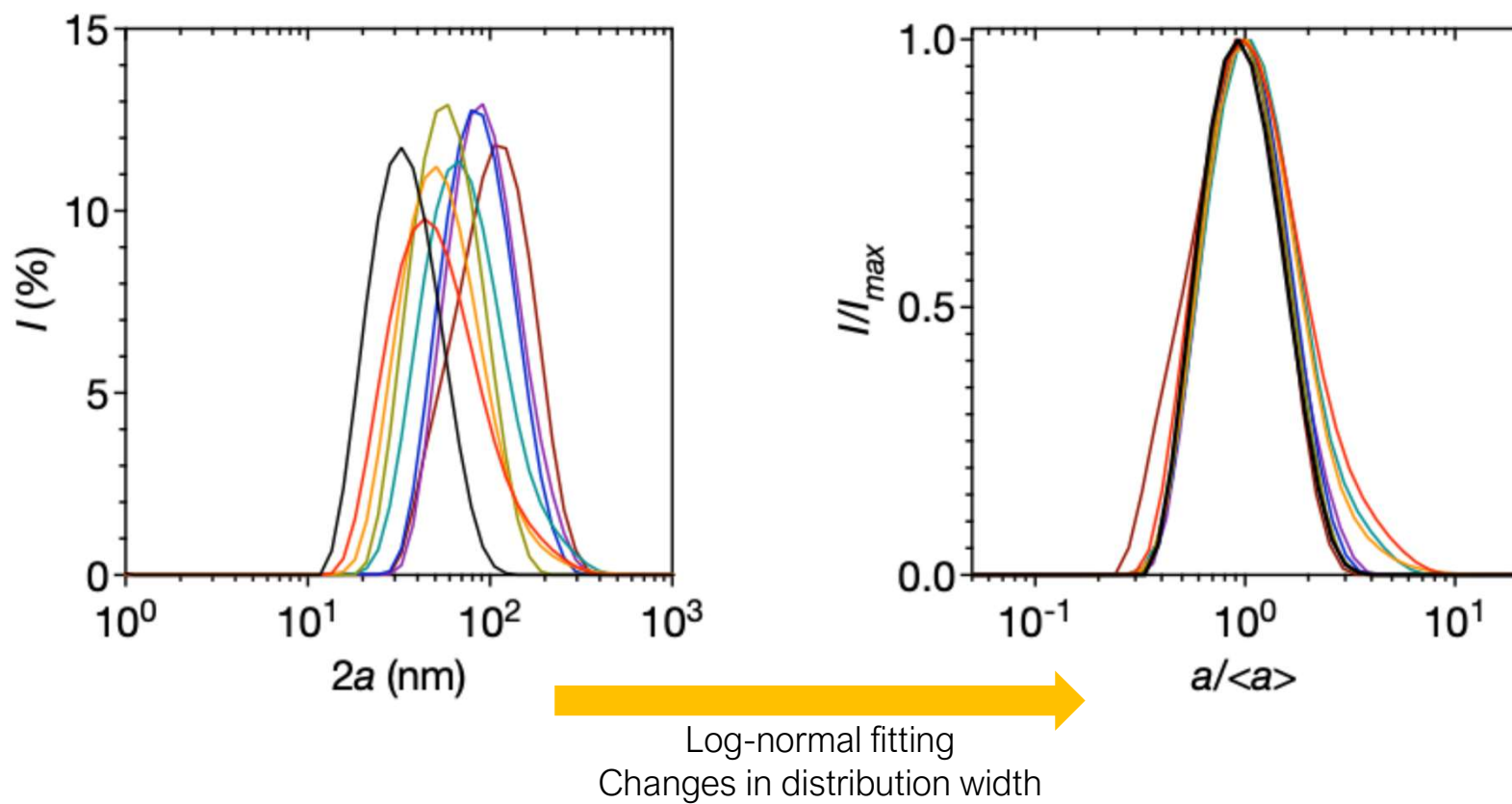
Coalescence



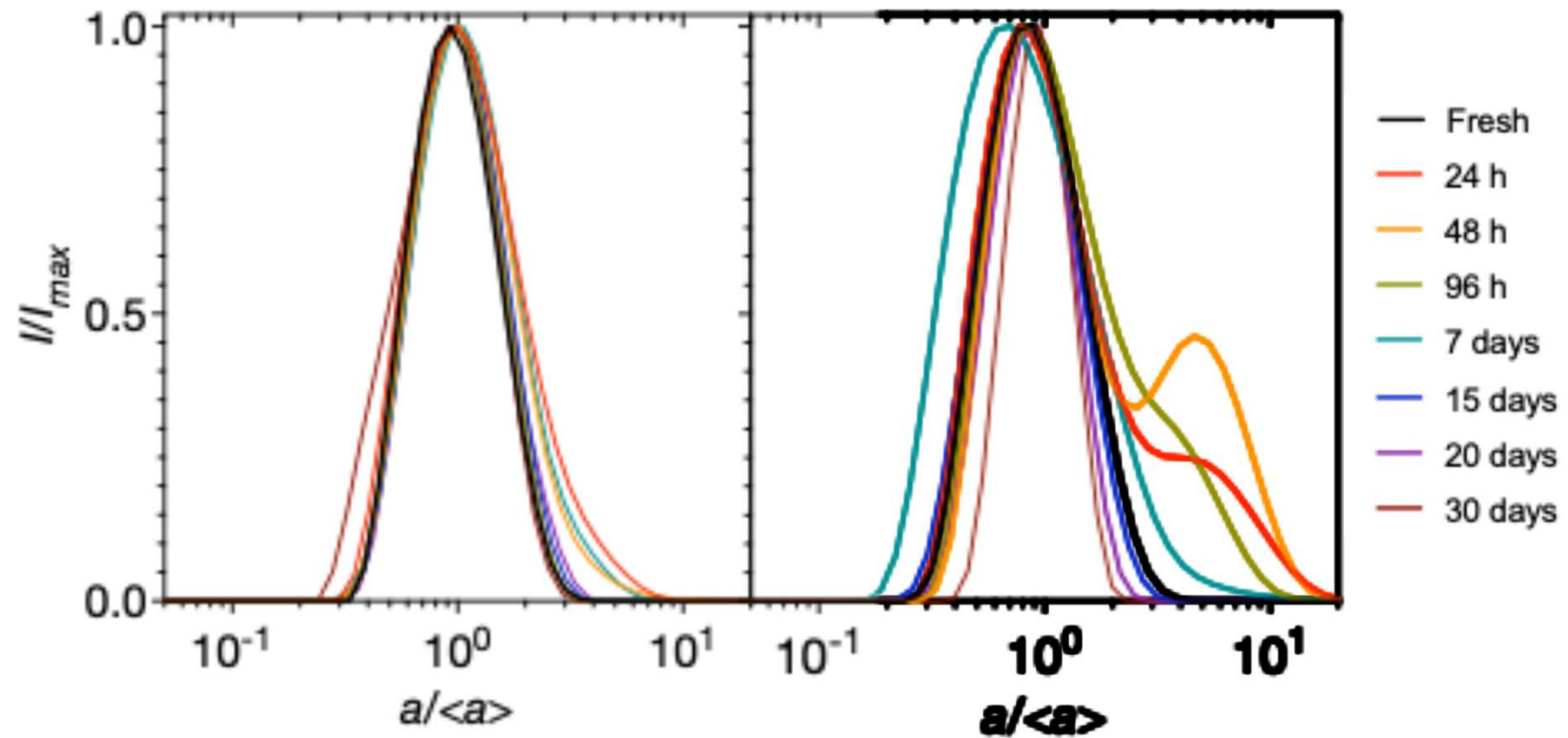
$$\langle r_H \rangle \sim t, t^2, t^{\frac{1}{3}}$$

Broadening PSD

Particle size distributions of coarsening emulsions

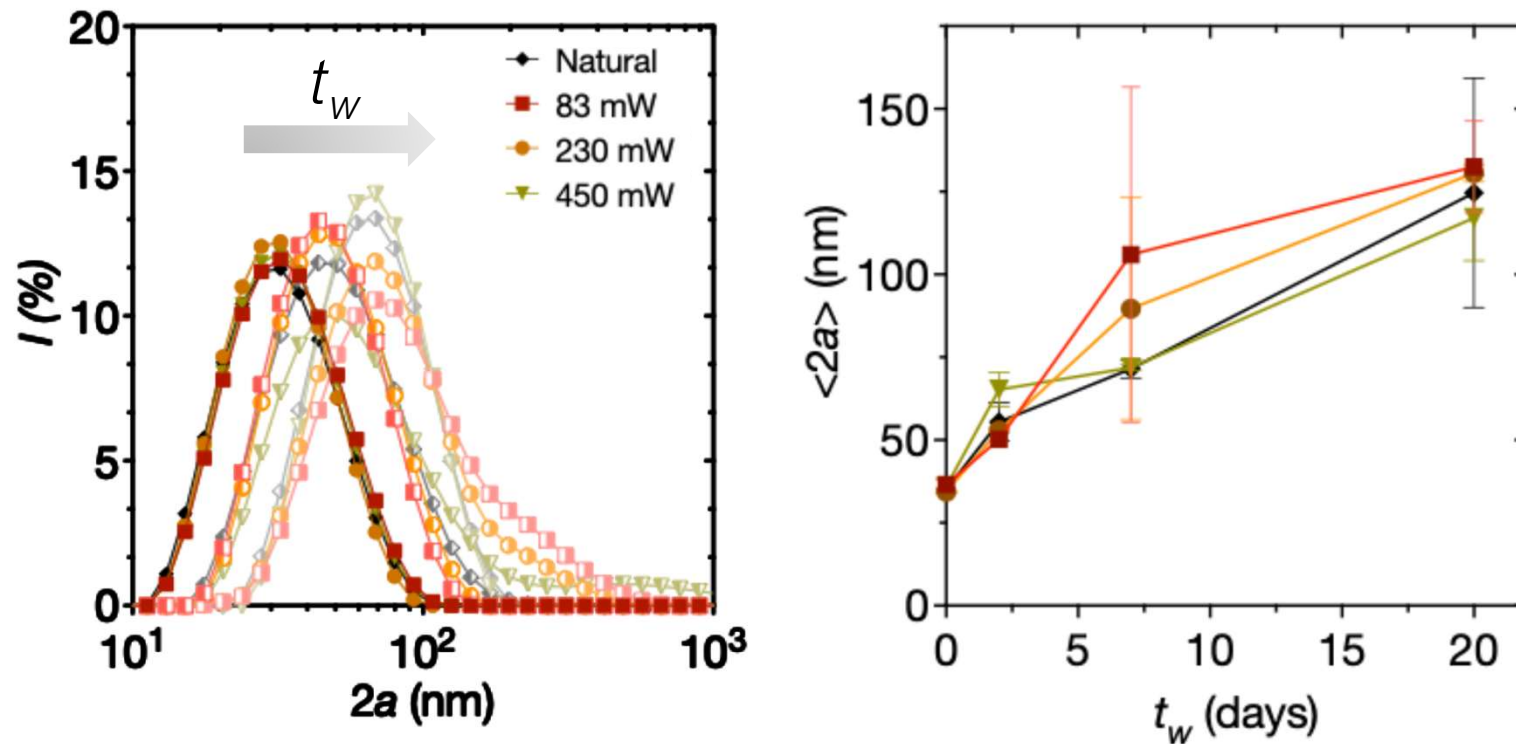


Effect of acoustics on particle size distribution



Bimodal distribution develops in acoustically aged samples
Generates onset of early-stage coalescence

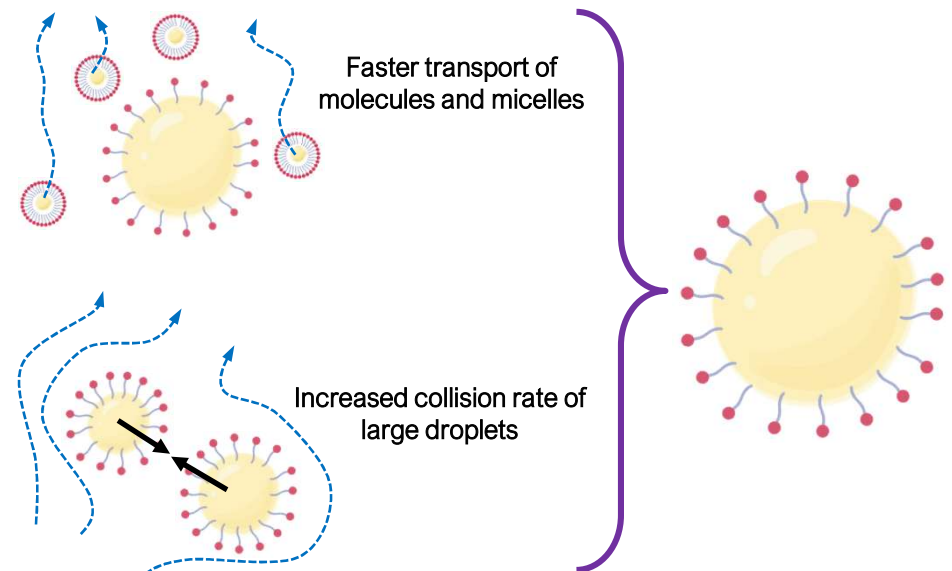
Effect of acoustic power on nanoemulsions



Increases coalescence rates with increasing P
 Little effect on the average diameter

Mechanism for accelerated coarsening in nanoemulsions

	Nanoemulsions
Attenuation/scattering/cavitation	-
Acoustic transmittance	0.8
Radiation pressure potential	$10^{-6} kT$
RP velocity ($\mu\text{m/s}$)	10^{-10}
Streaming velocity ($\mu\text{m/s}$)	1.3
Peclet number (particle)	0.01
Mechanism	Brownian/interfacial



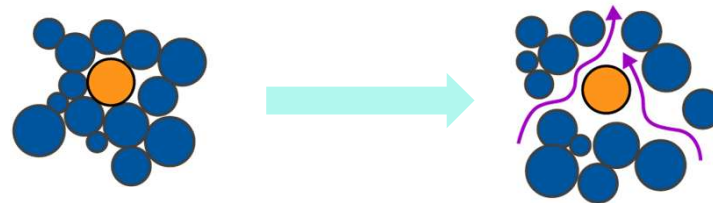
Streaming-induced Ostwald ripening/Marangoni flows

- Enhanced mass transport of barely soluble PDMS oil to droplets
- Increased surfactant gradient around individual droplets
- Negligible droplet or micelle deformation (Laplace pressures and timescales)

Summary and plans

Acoustics increase convective mass transport in wet systems

- Works very well in $\sim\mu\text{m}$ range
- Captures non-equilibrium processes such as vibrations during pumping/transportation
- Promising at enhancing **transport-limited processes such as mesoscale fluid drainage**
- Offers direct microscopic observability
- **Comparisons to accelerated syneresis could be fruitful (Del Gado project)**



Plans for Y4:

- *In situ* visualization of flow fields with acoustics
- Testing effect on syneretic systems (solvent separation from gel network)
- Systematic evaluation of acceleration factor in colloidal dispersions

Very special thanks to Brian Levy-Polis (FMC), Vicky Cheng (FMC), Kelly Krzysik (Dow), other members... and Eric Furst for asking if we figured it out yet



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