



# Experimental Investigations of Shear-Thickening and Yield Stress: The Roles of Friction and Surface Chemistry

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<sup>2</sup> ESPCI, Paris, France

<sup>3</sup> Monash University, Melbourne, Australia



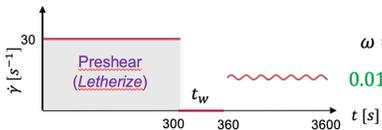
# Experimental Investigations of Shear-Thickening and Yield Stress: The Roles of Friction and Surface Chemistry

- **Part 1: Tuning** of the Shear-Thickening Fluid Transition
  - Anisotropic (highly rough) particles (long range interactions)
  - Surface chemistry (short range chemical interactions)
  - How low a volume fraction can we get to and still achieve discontinuous shear thickening (DST)?
  - Appearance of a yield stress in some of these colloidal dispersions
- **[Part 2]:** (Computational and Experimental) Micromechanics of the Yield Stress
  - Thixotropy and material aging
  - The key role of rolling friction
- **Part 3: Parallel Superposition** (combining steady and oscillatory shear) with Gabor transform for time/frequency separation to probe the properties of thixotropic elastoviscoplastic (TEVP) fluids
  - In a **thixotropic elasto-visco-plastic** (TEVP) material the stress typically evolves on two (often) distinct time-scales; **viscoelasticity** characterized by  $\tau_{ve}$  and **thixotropy**, characterized by  $\tau_{thix}$

Impose  $\gamma(t) = \dot{\gamma}_0 t + \gamma_0 \sin \omega t$   
 $\dot{\gamma} = \dot{\gamma}_0 + \dot{\gamma}_0 \cos \omega t$

$\omega = 1 \text{ or } 5 \text{ rad/s}$

$0.01 + 0.01 \omega (\cos \omega t)$

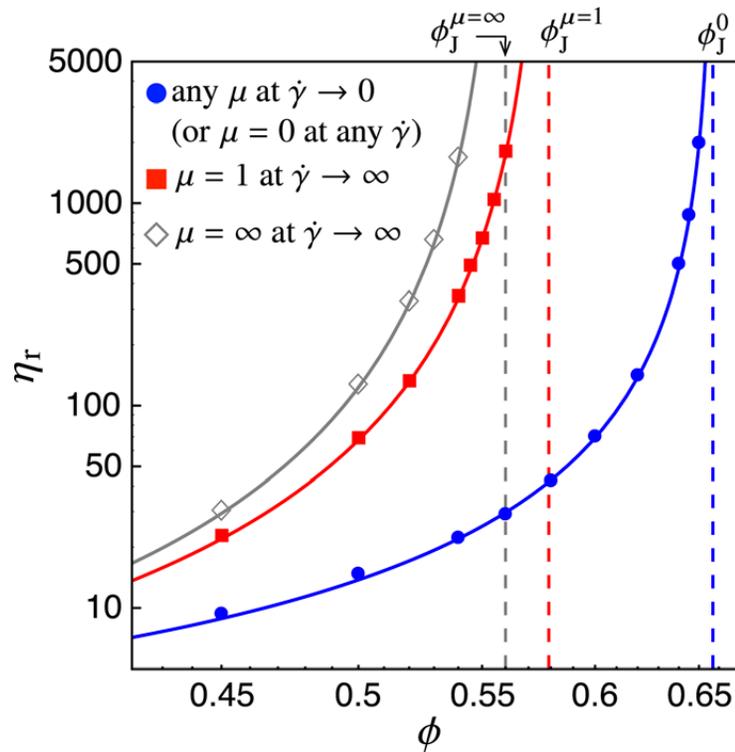


Monitor stress evolution in time:

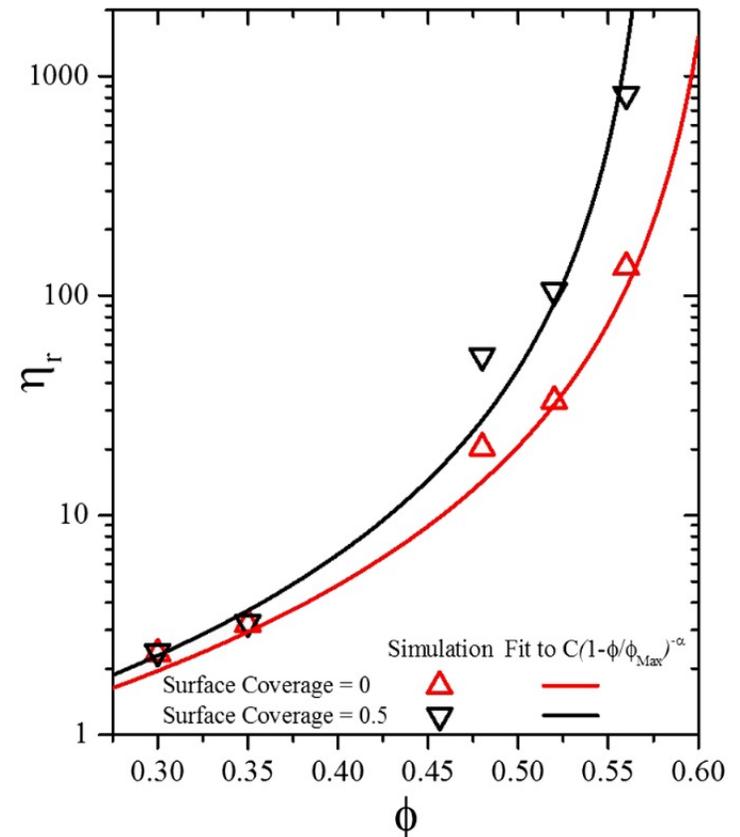
$$\sigma(t, \omega, \dot{\gamma}_0) = \eta_{eff}(t, \dot{\gamma}_0) \dot{\gamma}_0 + \gamma_0 \{ G'_{||}(\omega, t) \sin \omega t + G''_{||}(\omega, t) \cos \omega t \}$$

# The shear-thickening transition is set by **frictional interactions**

- Explore the competing roles of *roughness* ( $\mu$ ) and *surface chemistry* in controlling the shear thickening transition

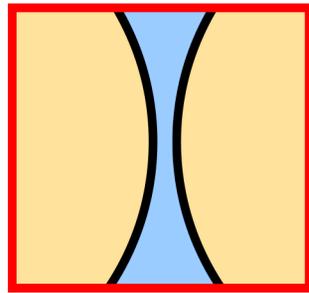
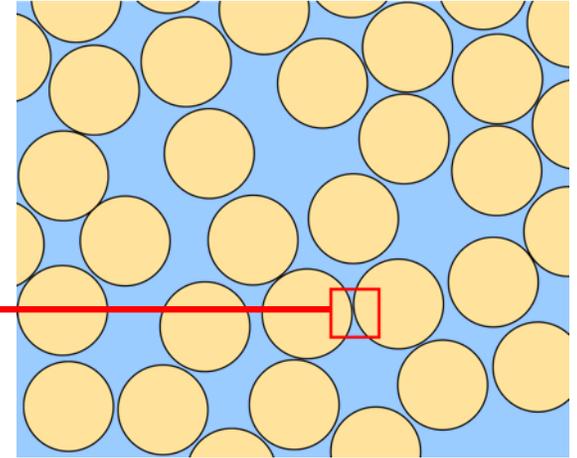
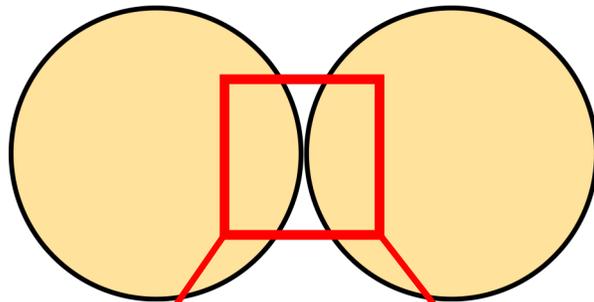


Mari, Seto, Morris & Denn, 2014  
 Boyer, Guazelli & Pouliquen, 2011  
 Wyart & Cates, 2014



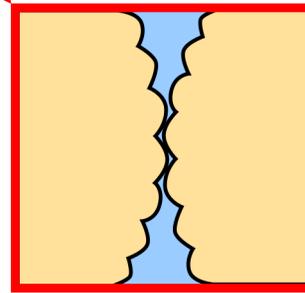
Jamali & Brady, 2019

# Frictional interactions



Hydrodynamic forces

Wagner *et al.*, 2009  
Lin *et al.*, 2015  
Jamali *et al.*, 2019



Solid friction

Wyart *et al.*, 2014  
Mari *et al.*, 2014  
Comtet *et al.*, 2015  
Hsu *et al.*, 2018

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## Tuning the shear thickening of suspensions through surface roughness and physico-chemical interactions

Philippe Bourriane <sup>1</sup>, Vincent Niggel <sup>1</sup>, Gatién Polly<sup>1,2</sup>, Thibaut Divoux <sup>2,3</sup> and Gareth H. McKinley <sup>1</sup>

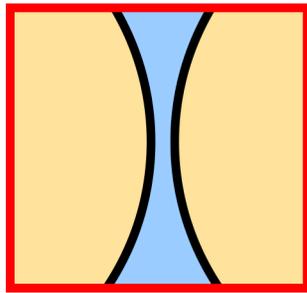
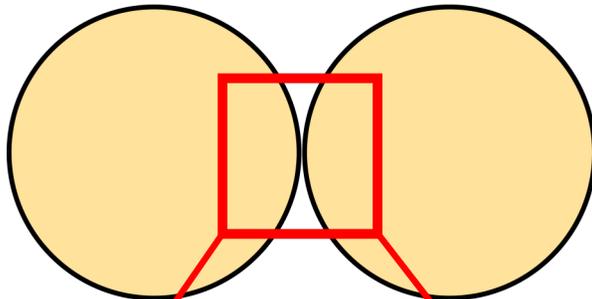
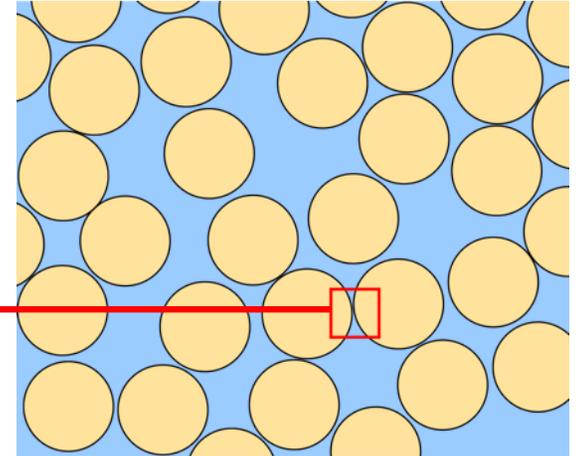
<sup>1</sup>Hatsopoulos Microfluids Laboratory, Department of Mechanical Engineering, MIT, 77 Massachusetts Avenue, Cambridge, Massachusetts 02139, USA

<sup>2</sup>MultiScale Material Science for Energy and Environment, UMI 3466, CNRS-MIT, 77 Massachusetts Avenue, Cambridge, Massachusetts 02139, USA

<sup>3</sup>ENSL, CNRS, Laboratoire de Physique, F-69342 Lyon, France

Bourriane *et al.*, *Phys Rev. Research* 2022

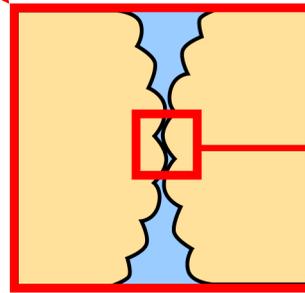
# Frictional interactions



Hydrodynamic forces

*Wagner et al., 2009*  
*Lin et al., 2015*  
*Jamali et al., 2019*

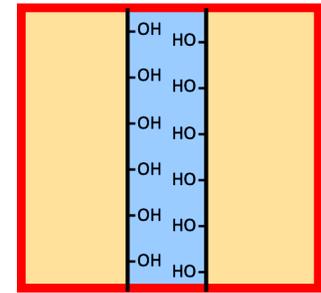
Roughness



Solid friction

*Wyart et al., 2014*  
*Mari et al., 2014*  
*Comtet et al., 2015*  
*Hsu et al., 2018*

Chemistry

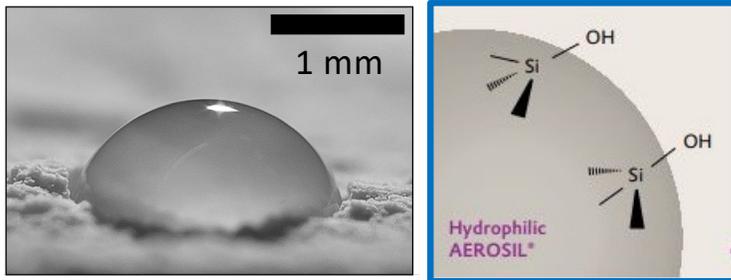


Hydrogen bonds

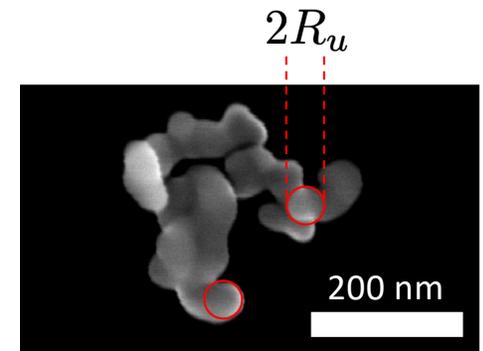
*Raghavan et al., 2000*  
*Comtet et al., 2015*  
*James et al., 2018*

# Suspensions of **fumed silica** in polypropylene glycol (polar solvent)

Raghavan *et al.*, 1995  
Raghavan *et al.*, 1997

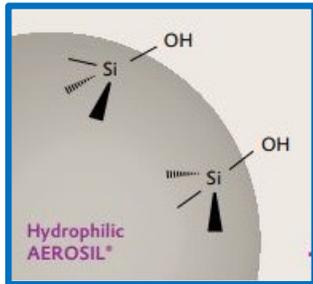
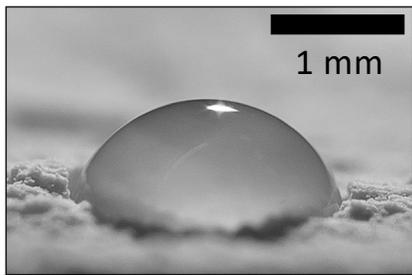


Hydrophilic (HP)

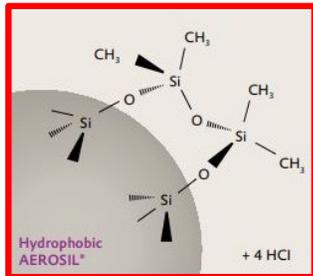
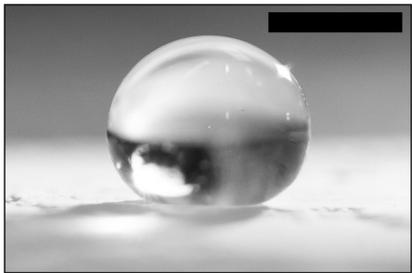


$$R_u = 25 \text{ nm}$$
$$\Sigma_s \sim 40 \text{ m}^2/\text{g}$$

# Suspensions of **fumed silica** in polypropylene glycol

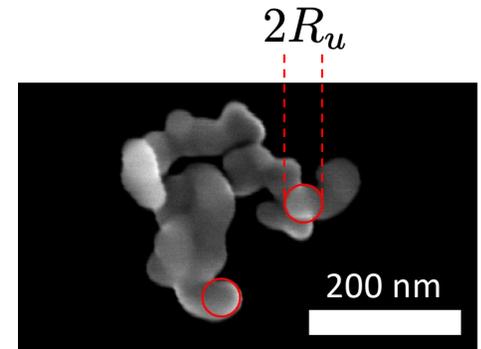


**Hydrophilic (HP)**



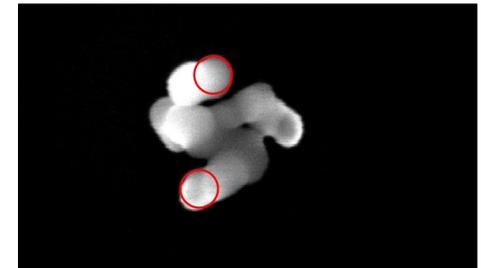
**Hydrophobic (HB)**

CHEMISTRY

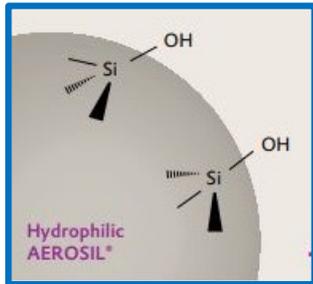
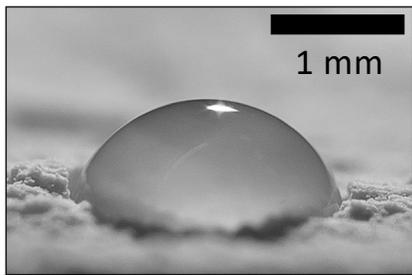


$$R_u = 25 \text{ nm}$$

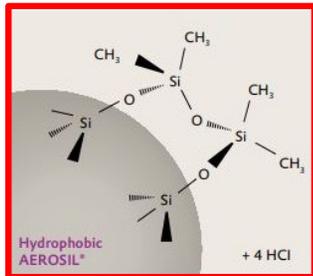
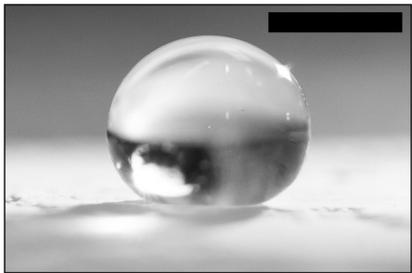
$$\Sigma_s \sim 40 \text{ m}^2/\text{g}$$



# Suspensions of **fumed silica** in polypropylene glycol

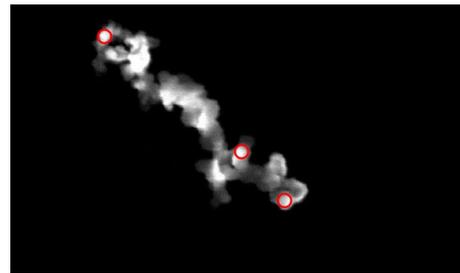


**Hydrophilic (HP)**



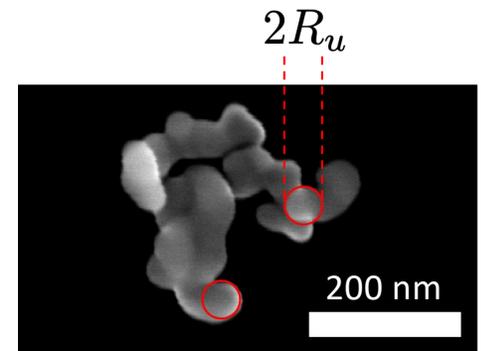
**Hydrophobic (HB)**

CHEMISTRY



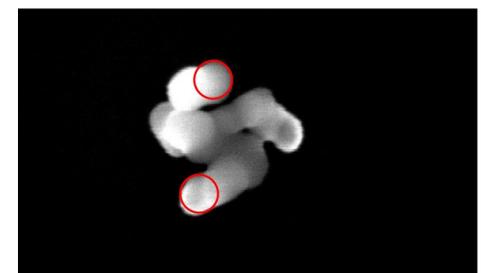
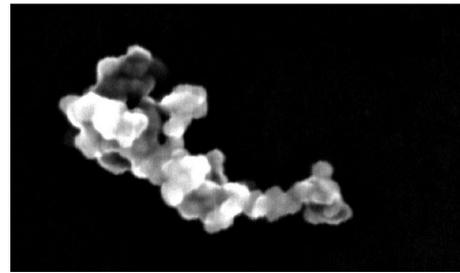
$$R_u = 10 \text{ nm}$$

$$\Sigma_s \sim 300 \text{ m}^2/\text{g}$$



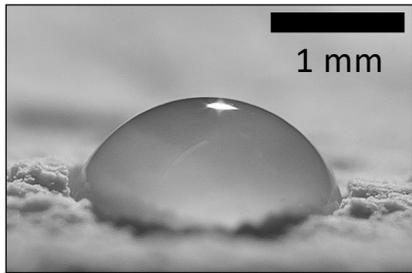
$$R_u = 25 \text{ nm}$$

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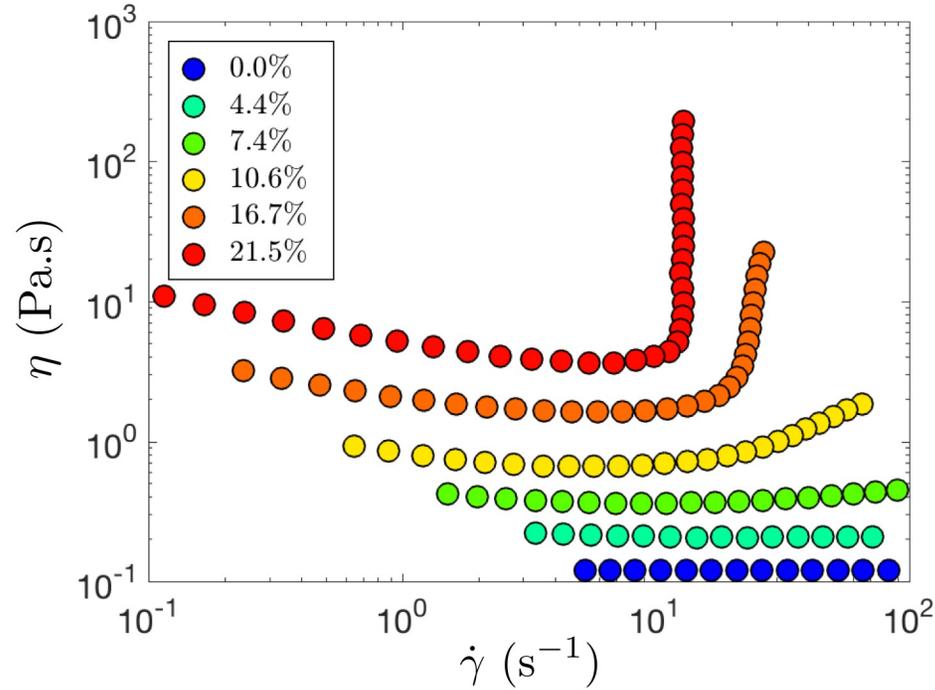


GEOMETRY – Nanometric roughness; high surface area/volume

# Shear-thickening transition at low volume fraction

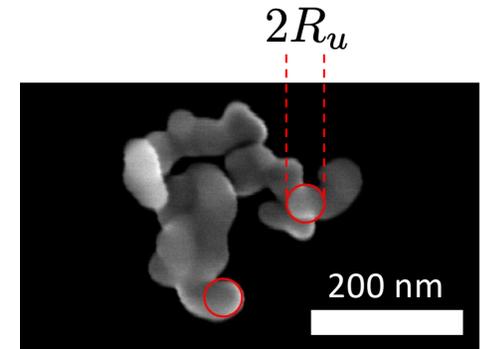


Hydrophilic (HP)



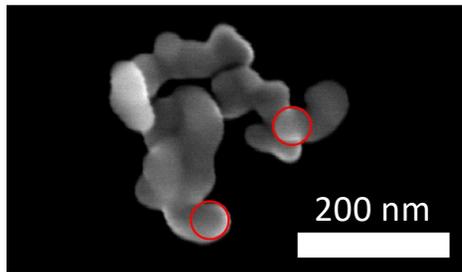
□ Continuous shear-thickening (CST)  
 $\phi > 5\%$

□ Discontinuous shear-thickening (DST)  
 $\phi > 17\%$



$$R_u = 25 \text{ nm}$$

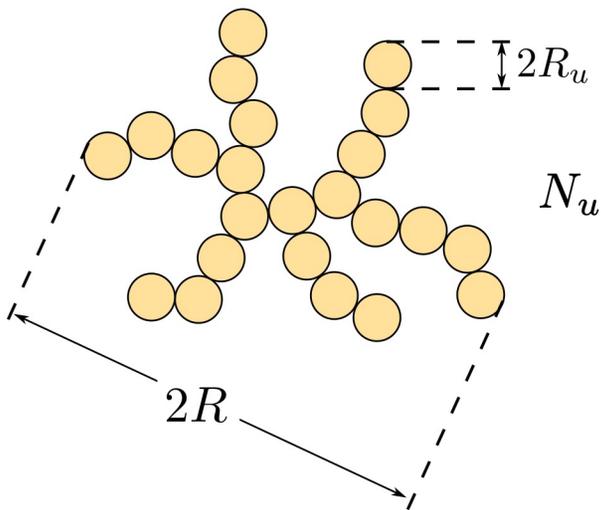
# Onset of shear-thickening



Interparticle distance  $L$

$$\phi = \frac{\Omega_p}{L^3}$$

$$\Omega_p = N_u \frac{4}{3} \pi R_u^3$$



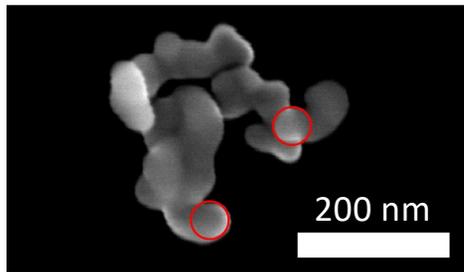
$$L = \left( \frac{4\pi R_u^3 N_u}{3\phi} \right)^{1/3}$$

$$R_u = 25 \text{ nm}$$

$$N_u \sim 25$$

$$R = 150 \text{ nm}$$

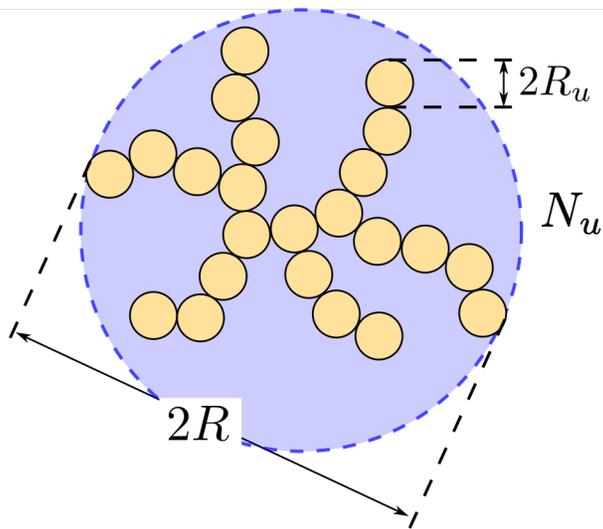
# Onset of shear-thickening



Interparticle distance  $L$

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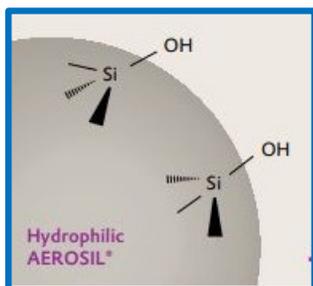
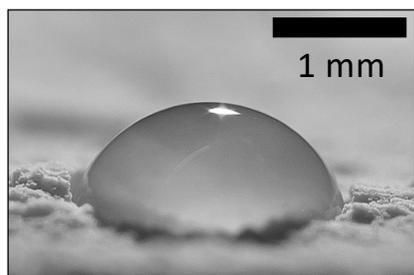
$$N_u \sim 25$$

$$R = 150 \text{ nm}$$

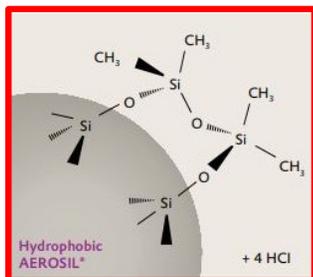
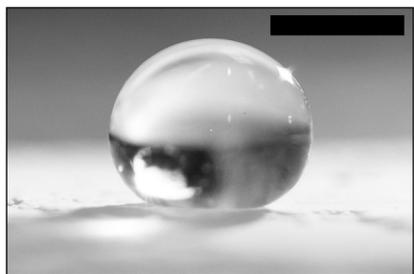
$$L(\phi^*) \sim 2R$$

$$\phi^* \sim 6\%$$

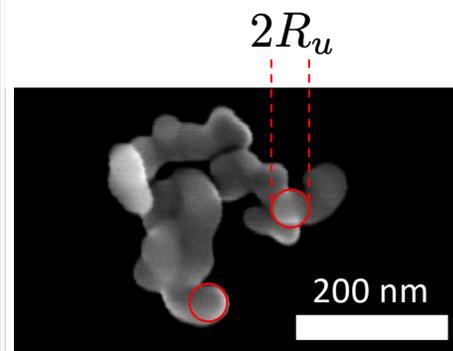
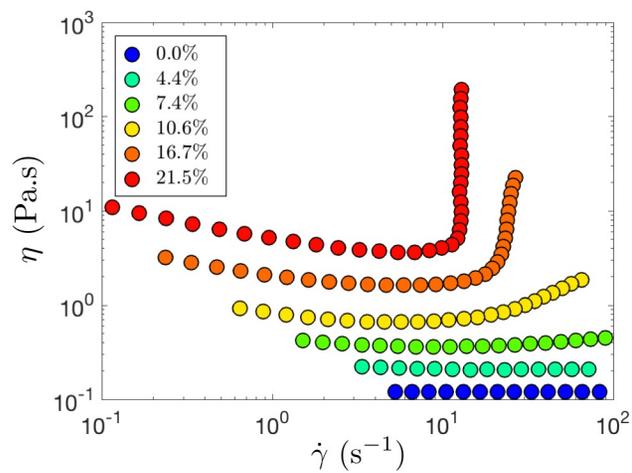
# Hydrophilic vs Hydrophobic



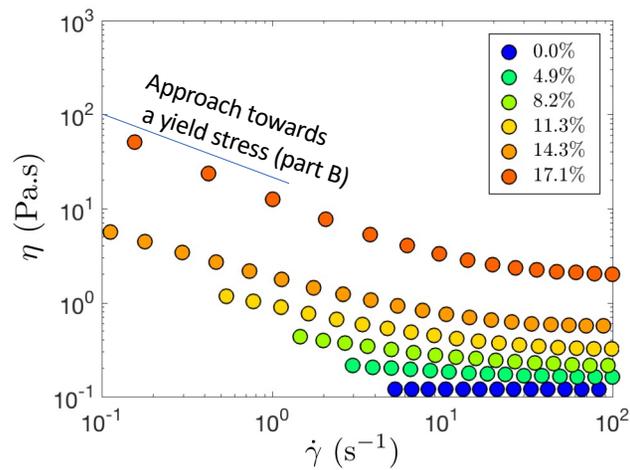
Hydrophilic (HP)



Hydrophobic (HB)

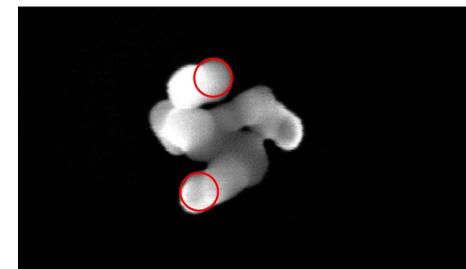


CHEMISTRY

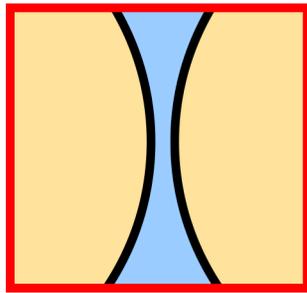
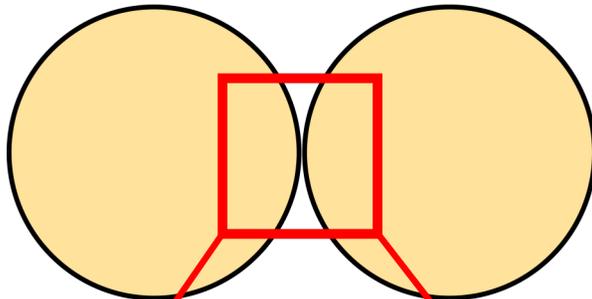
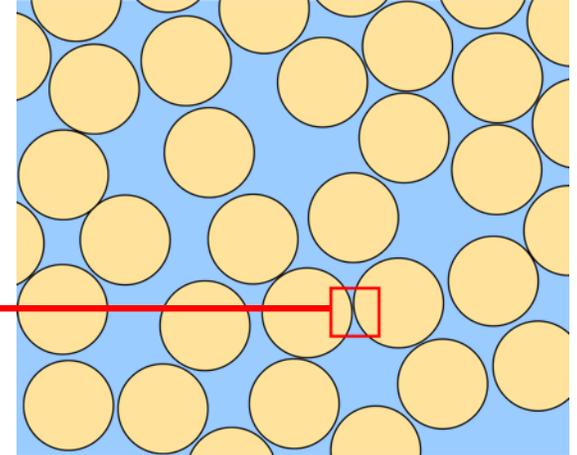


$$R_u = 25 \text{ nm}$$

$$\Sigma_s \sim 40 \text{ m}^2/\text{g}$$

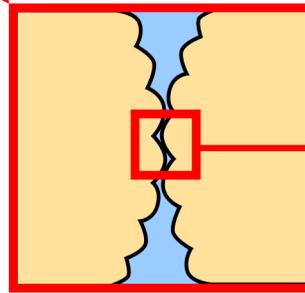


# Frictional interactions



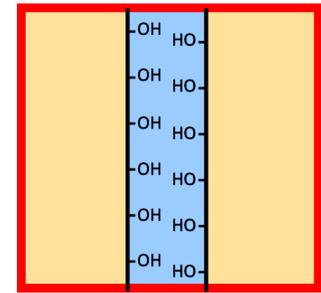
Hydrodynamic forces

## Roughness



Solid friction

## Chemistry



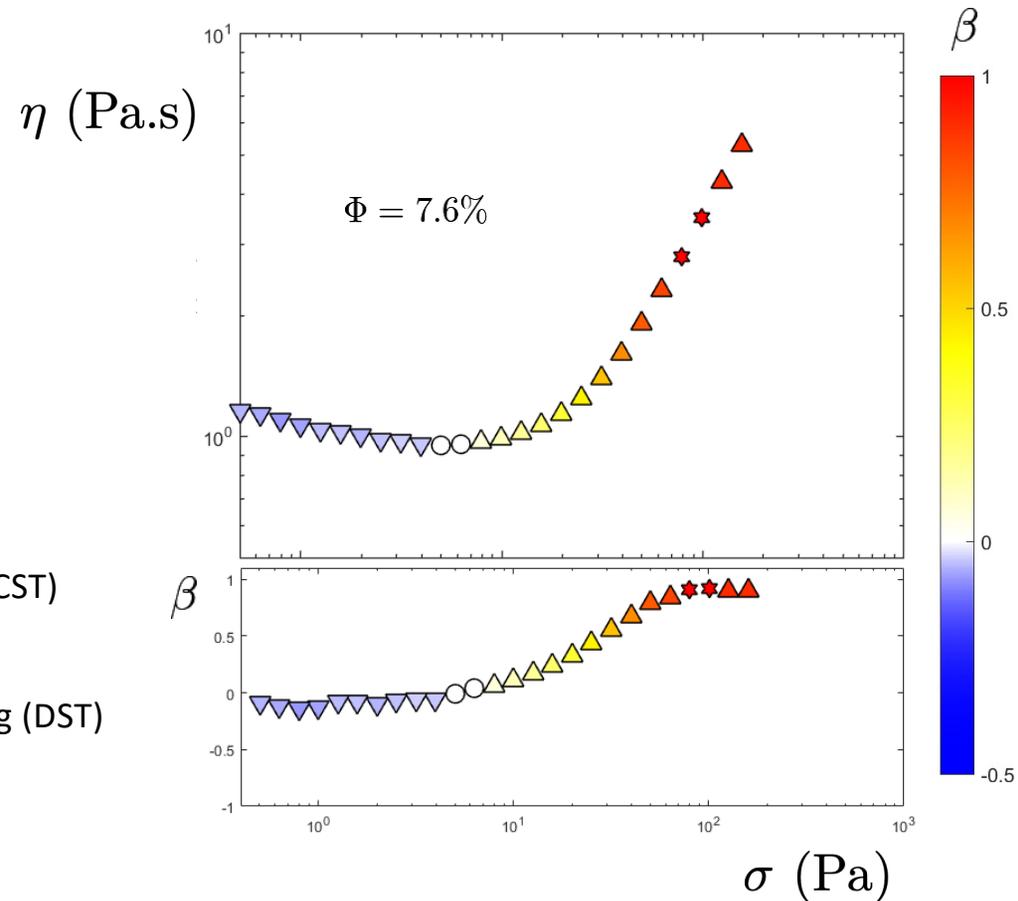
Hydrogen bonds

Hydrophilic vs Hydrophobic  
Key role on shear-thickening

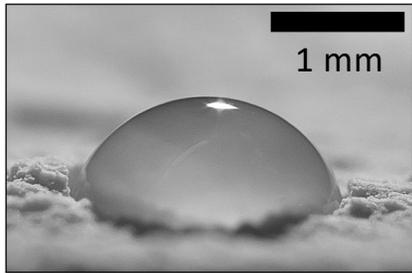
# Shear-thickening quantification

$$\beta = \dot{\gamma} \frac{d\eta}{d\sigma} = 1 - \eta \frac{d\dot{\gamma}}{d\sigma}$$

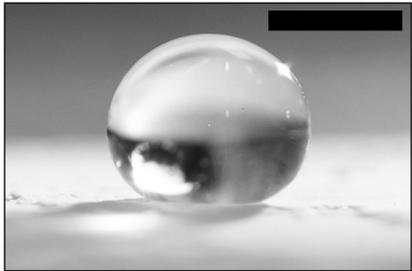
- $\beta \approx 0$   $\Rightarrow$  Newtonian
- ▽  $\beta < 0$   $\Rightarrow$  Shear thinning
- △  $0 < \beta < 1$   $\Rightarrow$  Continuous shear-thickening (CST)
- ★  $\beta = 1$   $\Rightarrow$  Discontinuous shear-thickening (DST)



# Influence of particle nanometric roughness

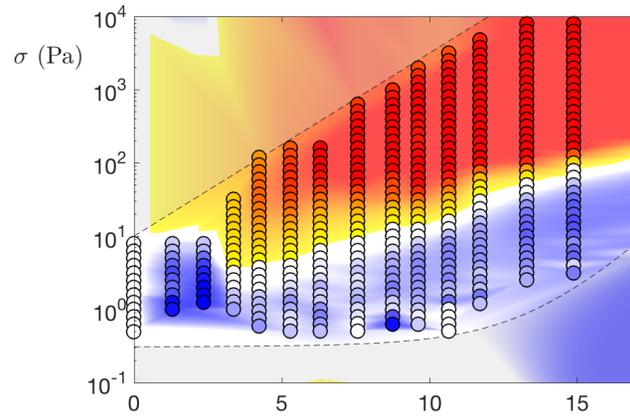


Hydrophilic (HP)

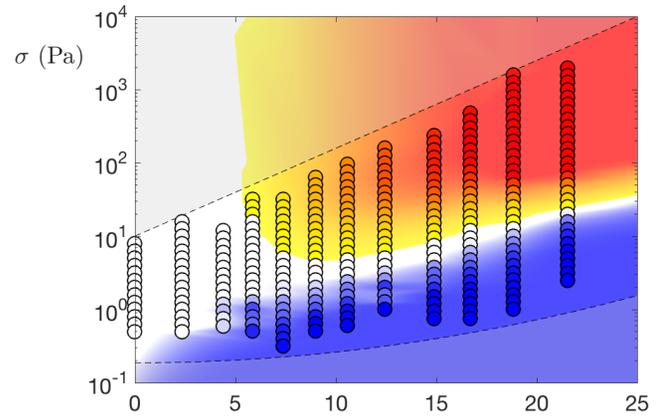
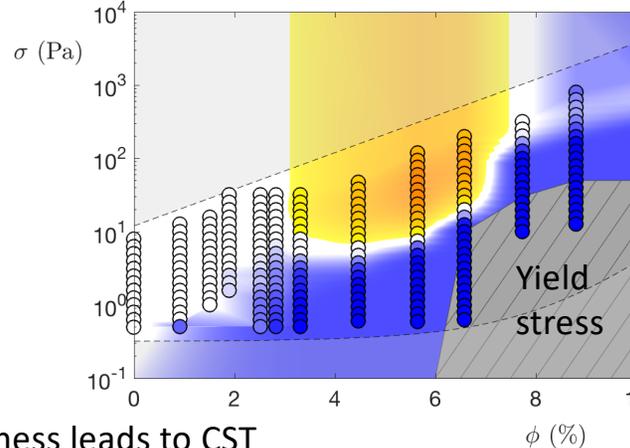


Hydrophobic (HB)

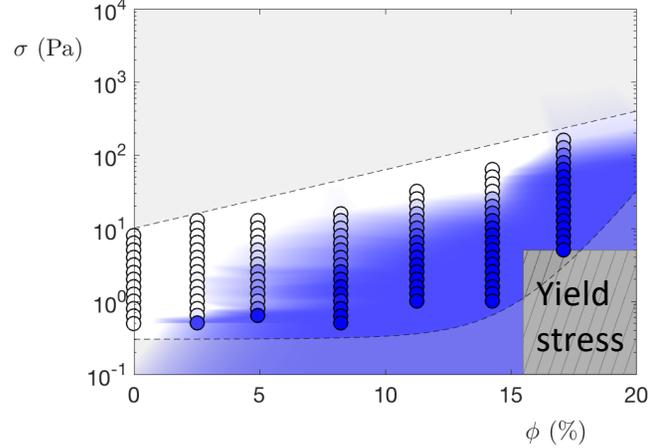
CHEMISTRY



$R_u = 10 \text{ nm}$   $\phi$  (%)  
 $\Sigma_s \sim 300 \text{ m}^2/\text{g}$



$R_u = 25 \text{ nm}$   $\phi$  (%)  
 $\Sigma_s \sim 40 \text{ m}^2/\text{g}$

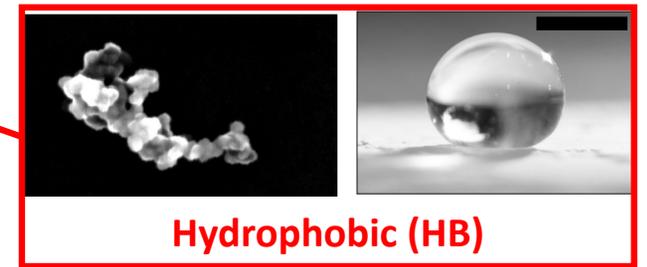
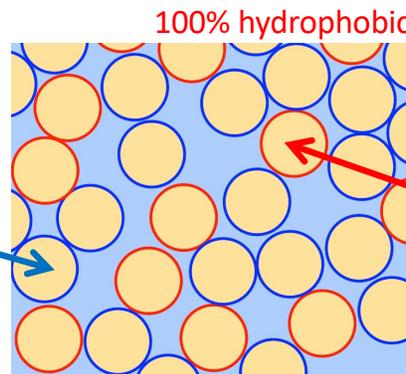
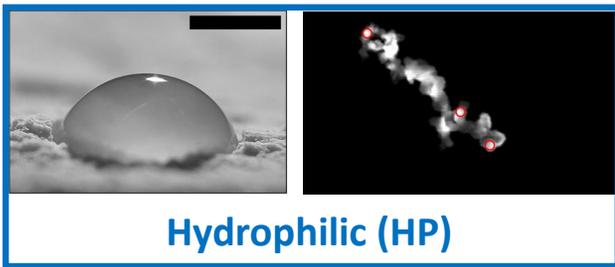
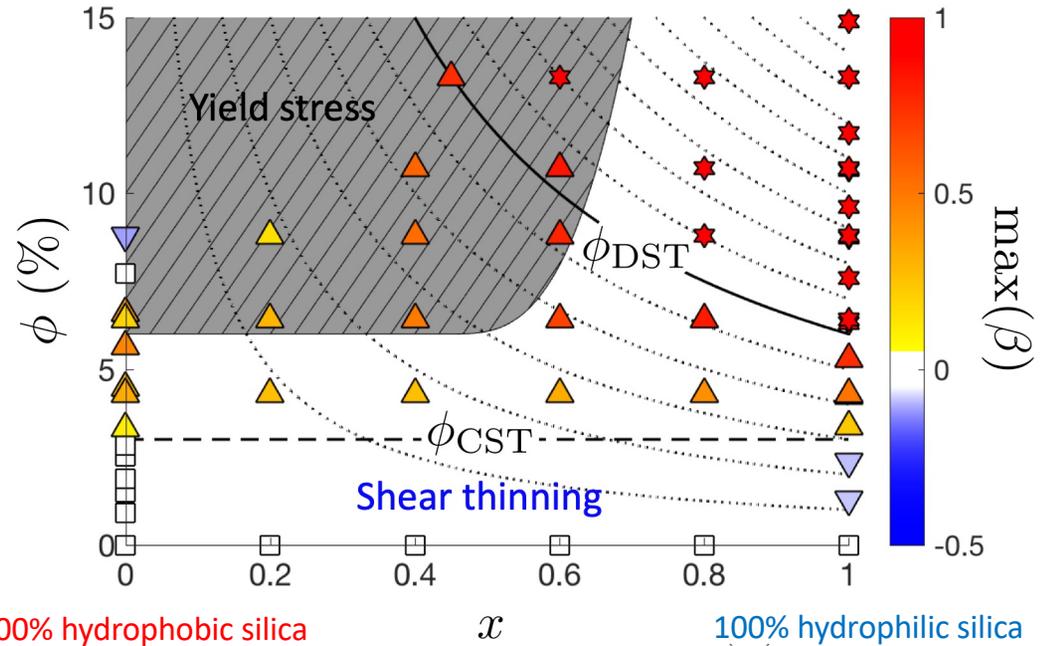
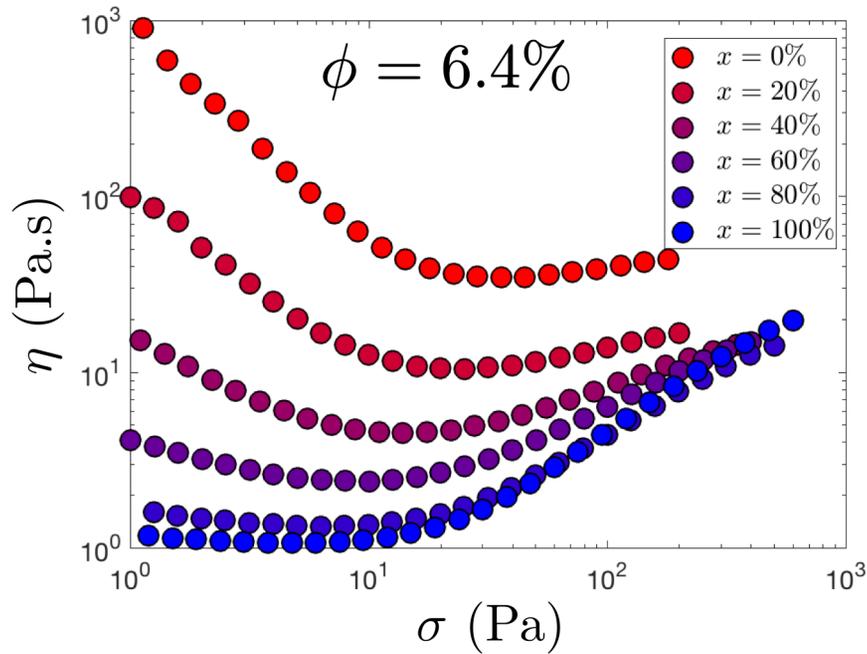


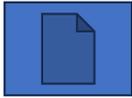
- Increasing nanometric roughness leads to CST in the absence of Hydrogen bonds

← GEOMETRY – Nanometric roughness

# Mixtures – tuning the shear-thickening response

- Constant volume fraction (and vary  $x = \text{hydrophilic fraction}$ )





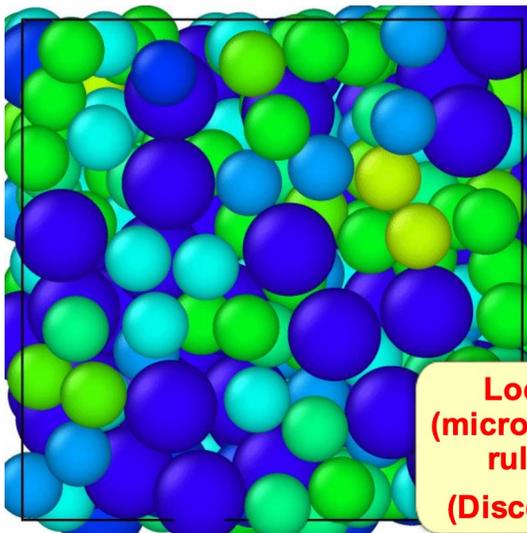
Simulations

# Micromechanics of Particulate Soft Matter



Experimental Methods (PSP)

## The Governing Role of Inter-Particle Interactions



Local  
(microscale)  
rules  
(Discover)



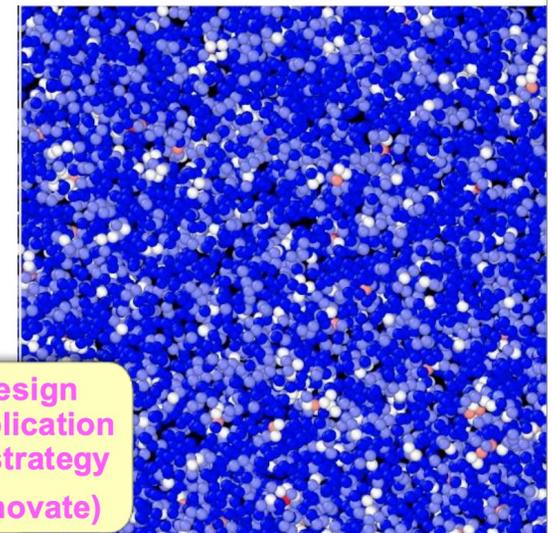
Latent  
(mesoscale)  
structure  
(Learn)



Global  
(macroscale)  
behavior  
(Predict)



Design  
(application  
strategy)  
(Innovate)



**Rishabh V. More, PhD**  
Department of Mechanical Engineering,  
Massachusetts Institute of Technology, Cambridge, MA USA  
(morer@mit.edu)



CompFlu-2023  
18-20 Dec 2023



MIT Portugal



Pharos Materials, Inc.  
Front Electrode Silver Paste Manufacturer



U.S. DEPARTMENT OF  
**ENERGY**

# Evolving rheology of thixotropic wet suspension: clay-based drilling mud



(an aging TEVP material)

Time-Dependent Lissajous Curves (LAOStress)

Time [seconds]  
(Aging)

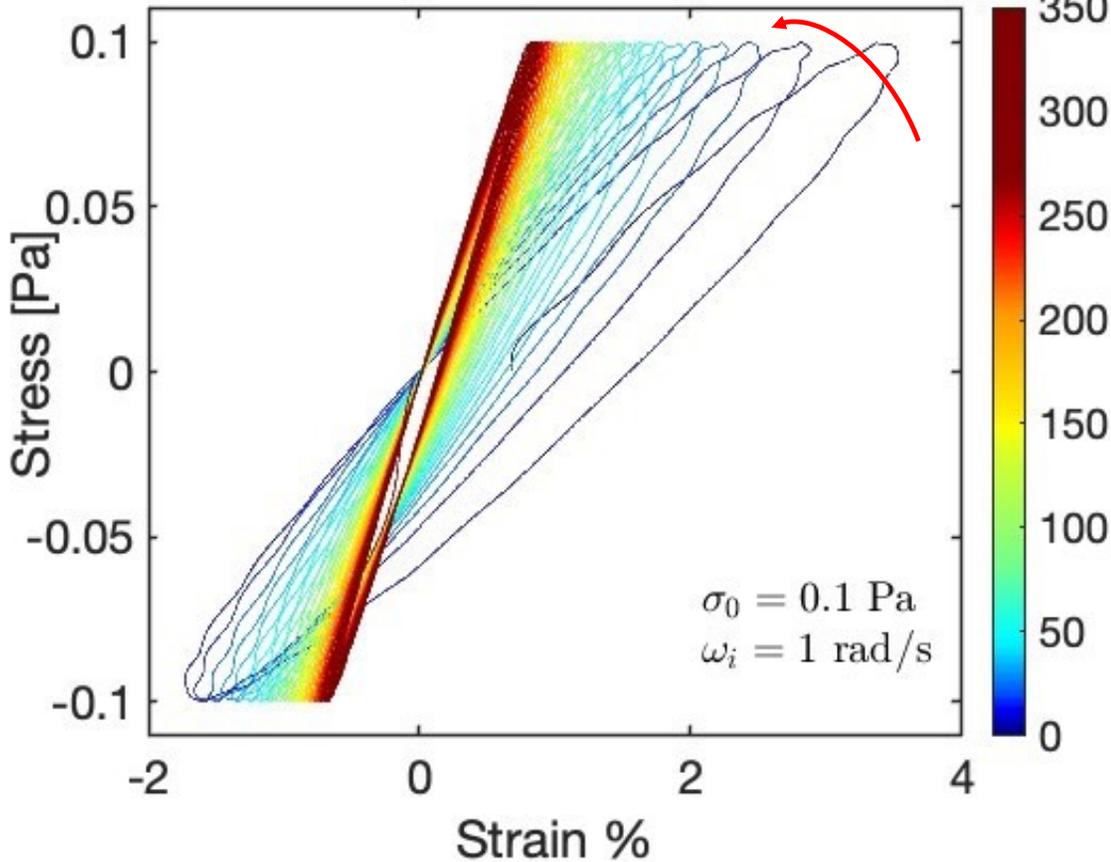
Material: Bentonite;  
Concentration: 5 wt.%;  
Temperature: 25°C  
Geometry: Cone 60 mm, 2°  
Instrument: DHR  
Pre-shear Protocol:  $500 \text{ s}^{-1}$  for 30 s  
Wait time after pre-shear ( $t_w$ ): 10 s.

SAOS Input:  $\sigma = \sigma_0 \sin(\omega_0 t)$

$\sigma_0 = 0.1 \text{ Pa}$   
 $\omega_0 = 1 \text{ rad/s}$



5wt% Bentonite dispersed in water



# Application of The Gabor Transform in Rheometric Measurements: *Gaborheometry*

- The Gabor transform is a special case of the Short Time Fourier Transform (STFT) in which the window function is a **Gaussian**.

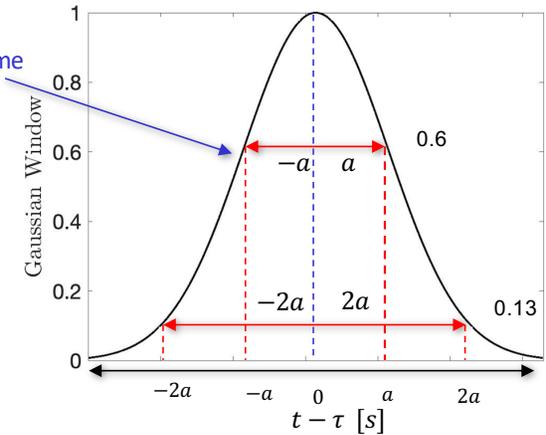


Dennis Gabor

Gabor Transform  $X(\omega, \tau) = \int_{-\infty}^{\infty} x(t) \underline{g(t - \tau)} e^{-i\omega t} dt$

Gaussian Window  $\underline{g(t - \tau)} = \frac{1}{\sqrt{2\pi} a} \exp\left(-\frac{(t - \tau)^2}{2a^2}\right)$

$a$  has units of time



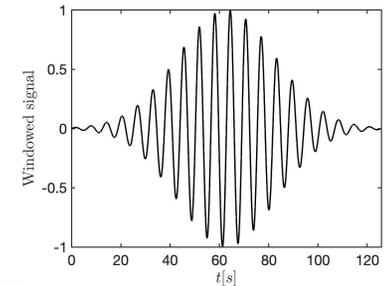
- In rheometry we care about getting amplitudes (moduli) right:**

For a sinusoidal signal  $x(t) = A \sin(\omega_0 t)$ , the power spectrum is given by:

$$|X(\omega, \tau = 0)| = \left| \frac{A}{2} e^{-\frac{a^2}{2}(\omega - \omega_0)^2} - \frac{A}{2} e^{-\frac{a^2}{2}(\omega + \omega_0)^2} \right|$$

- For the Gabor transform to be accurate, the amplitude of the computed spectral coefficient needs to be  $A/2$  at  $\omega = \omega_0$  and also at  $\omega = -\omega_0$ .

At  $\omega = \omega_0$  an **amplitude error** is incurred from the windowing:  $|X(\omega, \tau = 0)| = \frac{A}{2} - \frac{A}{2} e^{-\frac{a^2}{2}(2\omega_0)^2}$



- Getting the amplitude wrong from the Fourier spectrum means getting the modulus wrong!

...Evidently if the window length tends to infinity, we recover correct amplitude  $\frac{A}{2}$ .

*Heisenberg's uncertainty principle in rheometry!*

$$a \geq \frac{2.63}{\omega_0}$$

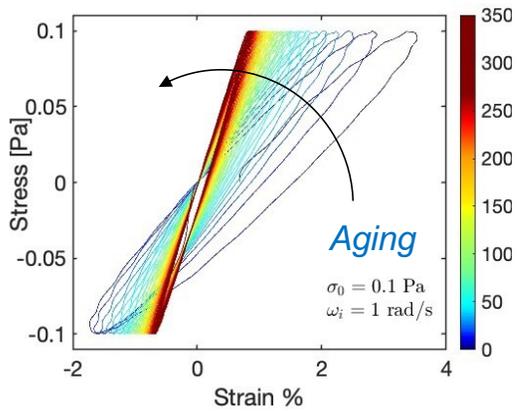
# Extracting $G'(\omega_0, t)$ & $G''(\omega_0, t)$ for an aging Bentonite clay (an aging TEVP material)



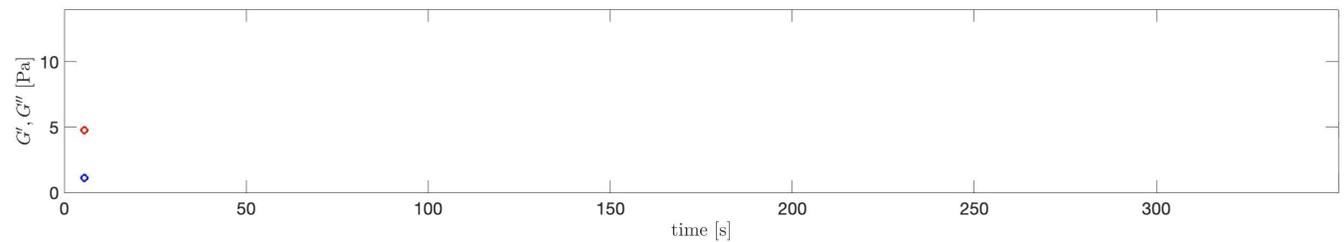
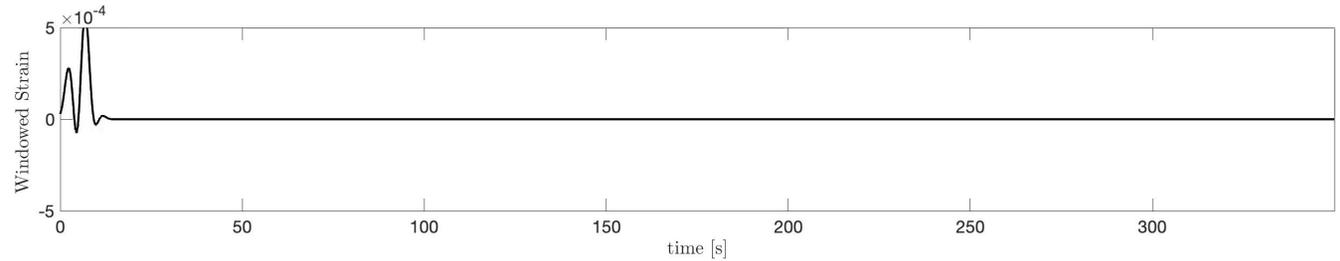
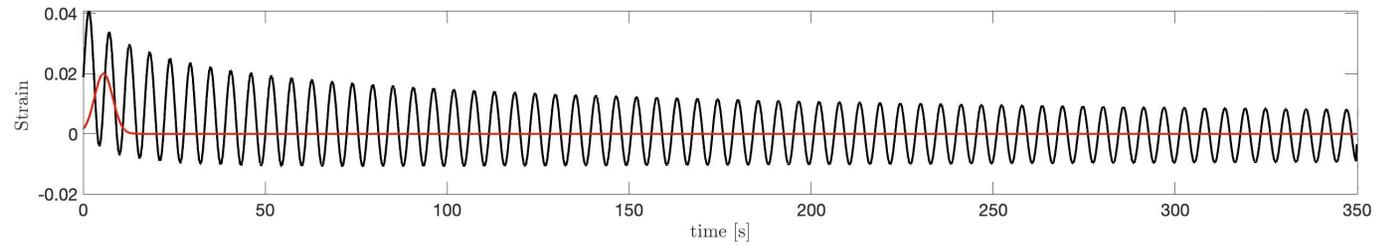
**Material:** Bentonite;  
**Concentration:** 5 wt.%;  
**Temperature:** 25°C  
**Geometry:** Cone 60 mm, 2°  
**Instrument:** DHR  
**Pre-shear Protocol:**  $500 \text{ s}^{-1}$  for 30 s  
**Wait time after pre-shear ( $t_w$ ):** 10 s.

**Input:**  $\sigma = \sigma_0 \sin(\omega_0 t)$   
 $\sigma_0 = 0.1 \text{ Pa}$   
 $\omega_0 = 1 \text{ rad/s}$

Thixotropic Lissajous Curve



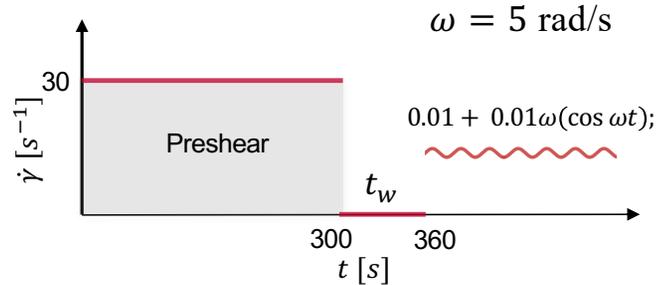
● Loss modulus      ○ Storage modulus       $a\omega_0 = 2.6$



Continuous logarithmic aging in time  $G'(\omega, t_{age}) \approx G'(\omega) * \ln\left(\frac{t_{age}}{\tau_{thix}}\right)$



# Compare with a Viscoelastic Polymer Solution (3wt% high MW PIB in Paraffin)



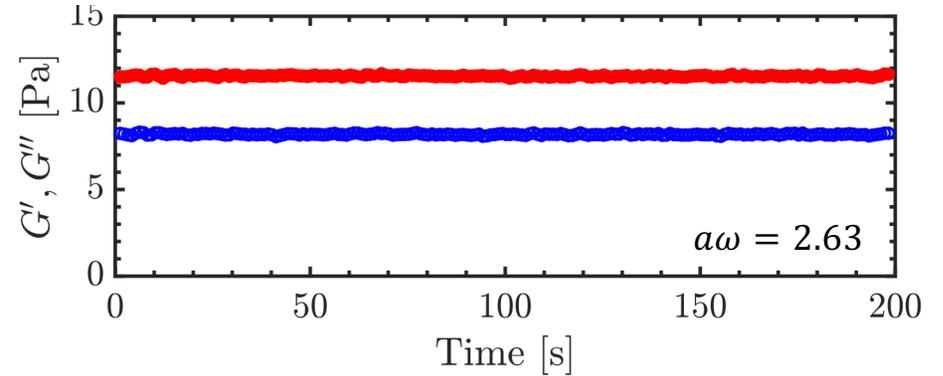
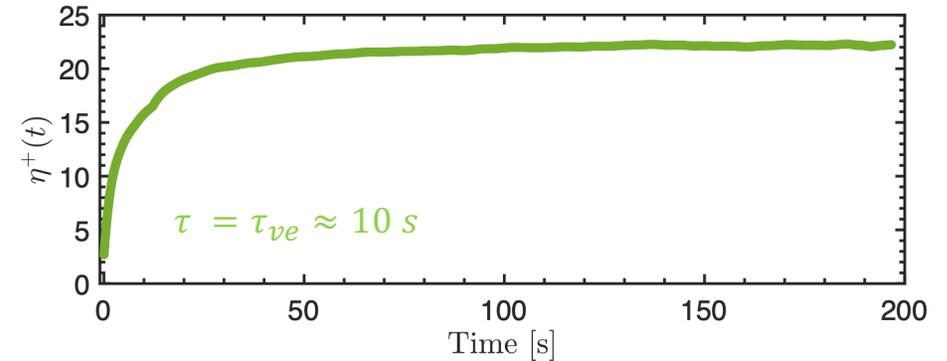
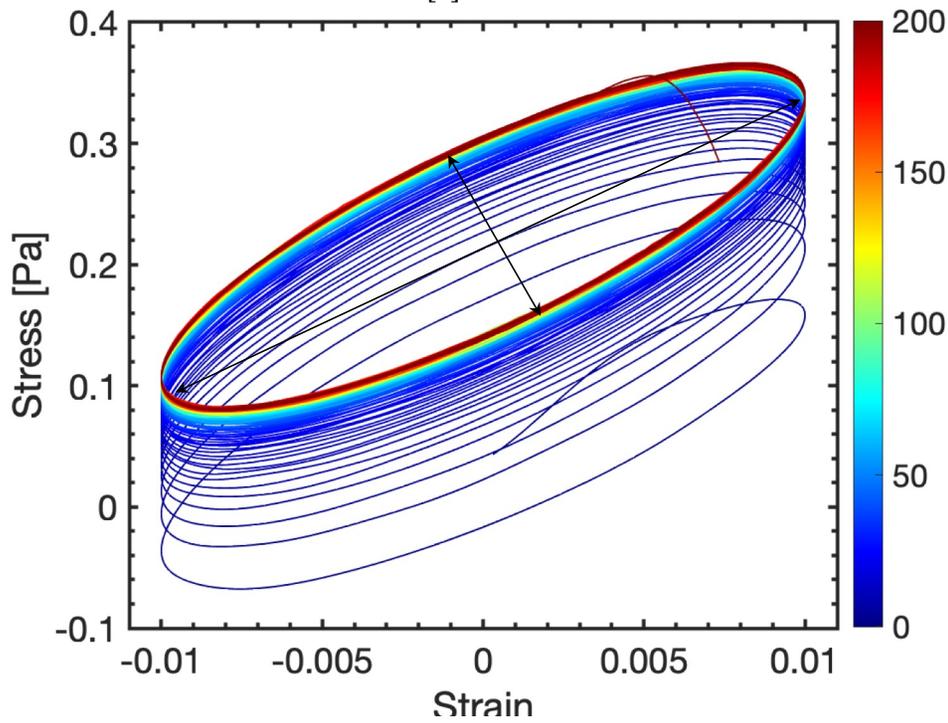
- Monitor the oscillating stress evolution in time:

$$\sigma(t, \omega, \dot{\gamma}_0) = \eta^+(t, \dot{\gamma}_0) \dot{\gamma}_0 + \gamma_0 \{ G'_{||}(\omega, t) \sin \omega t + G''_{||}(\omega, t) \cos \omega t \}$$

Evolution of DC term

Periodic evolution of viscoelastic response

- Phase angle of the Lissajous ellipse *does not change in time*



# Gabor-PSP rheometry for a silica dispersion at a high shear rate ( $Wi \gg 1$ )

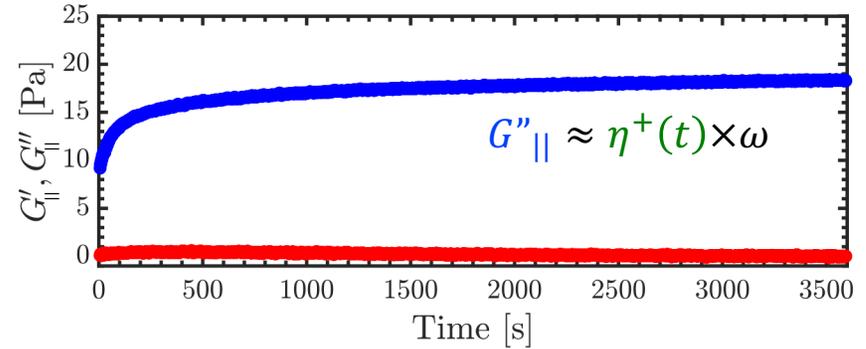
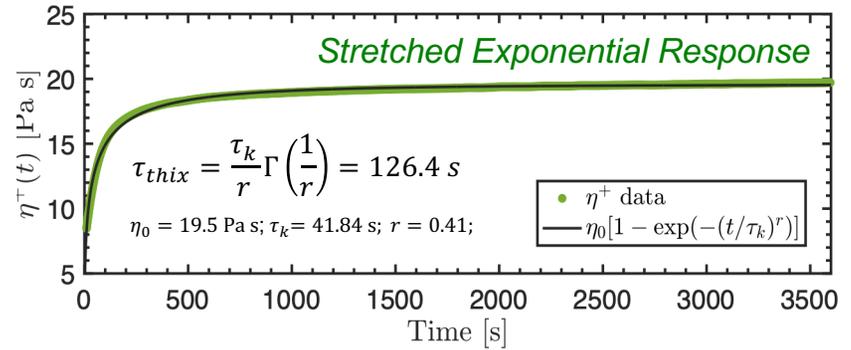
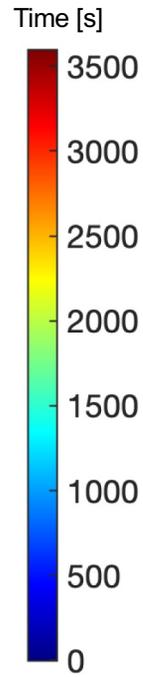
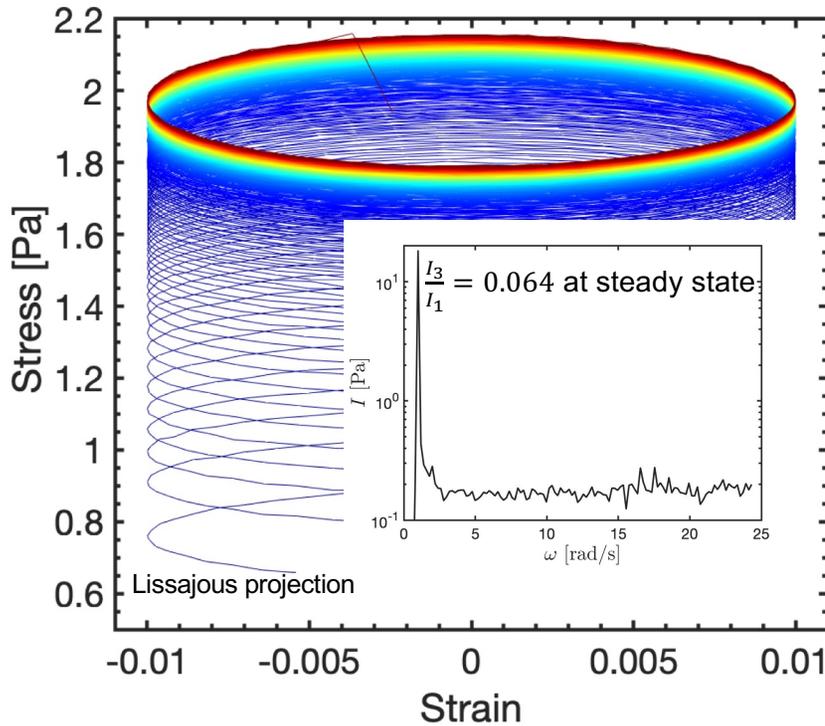
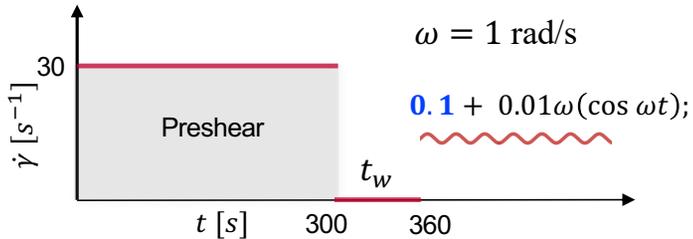


- Shear-melt the viscoelastic microstructure at high  $\dot{\gamma}_0$

$$\sigma(t, \omega, \dot{\gamma}_0) = \eta^+(t, \dot{\gamma}_0) \dot{\gamma}_0 + \gamma_0 \{ G'_{||}(\omega, t) \sin \omega t + G''_{||}(\omega, t) \cos \omega t \}$$

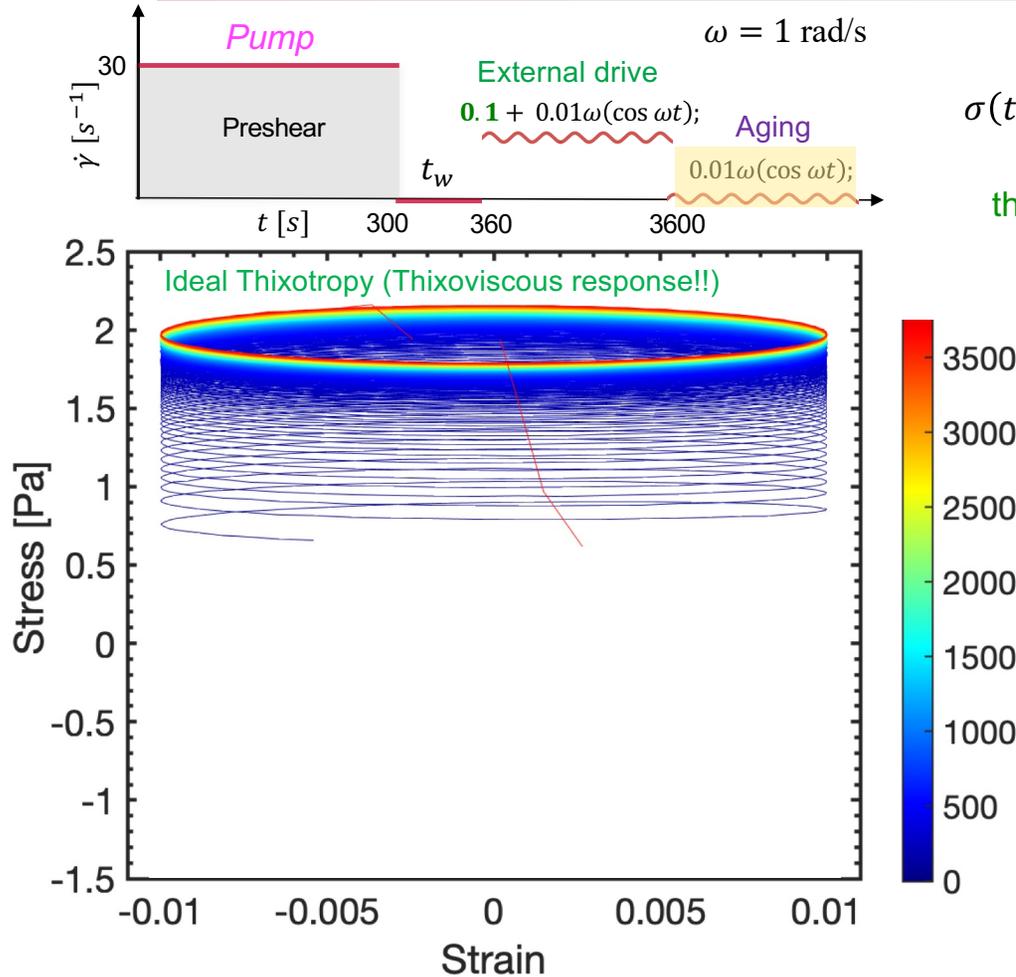
“slow” (inelastic) thixotropic evolution of DC term

“fast” and periodic evolution of viscous response



**INELASTIC:** Now the viscous response truly mirrors the thixotropic response; **IDEAL THIXOTROPY**

...so finally...a clear rheometric way to systematically distinguish *Thixotropy* from *Aging*

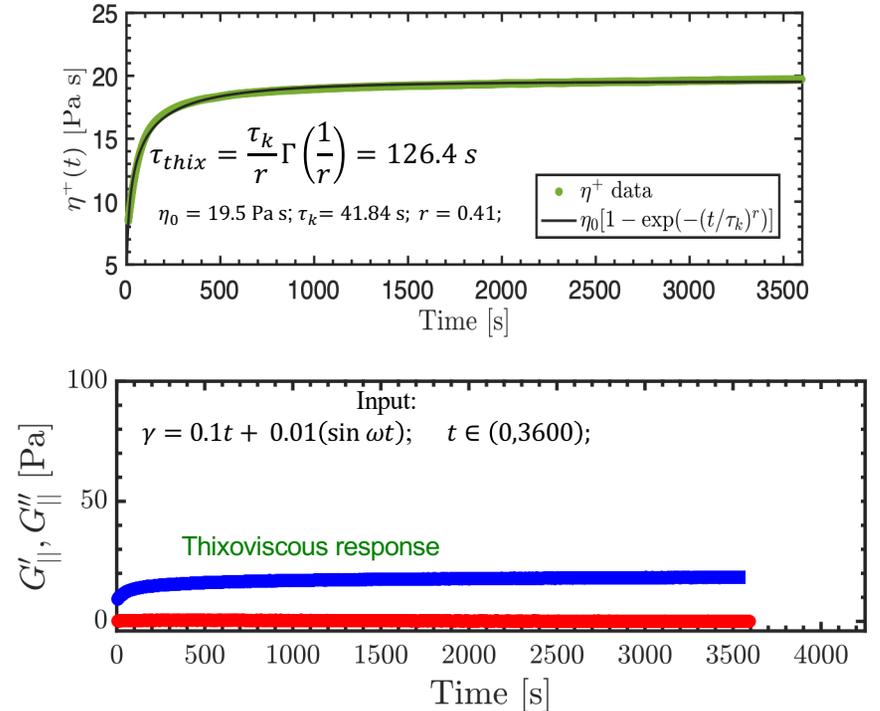


2.5 wt% Colloidal Silica Gel

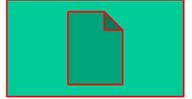
Monitor the oscillating stress evolution in time:

$$\sigma(t, \omega, \dot{\gamma}_0) = \eta^+(t, \dot{\gamma}_0) \dot{\gamma}_0 + \gamma_0 \{ \dots + G''_{||}(\omega, t) \cos \omega t \}$$

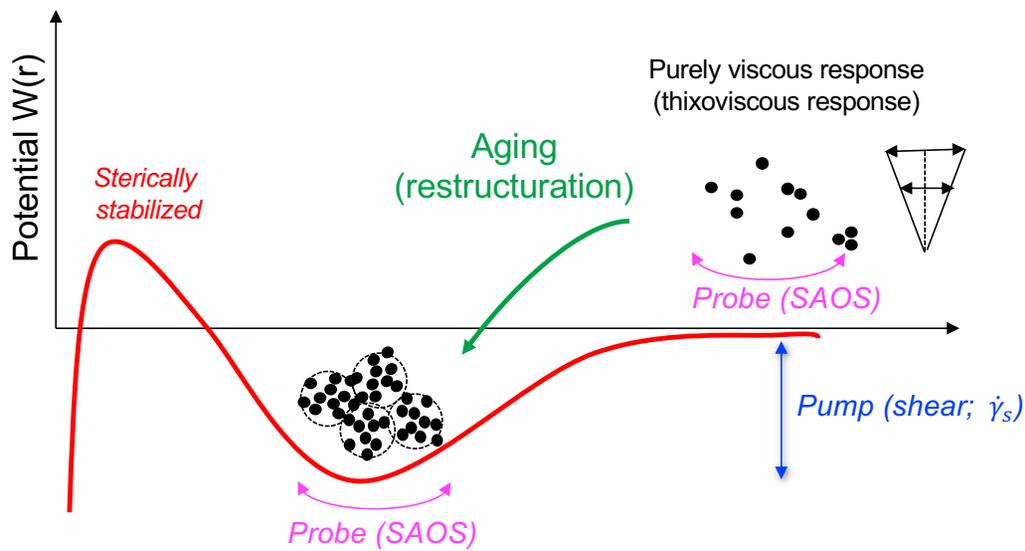
“slow” (inelastic) thixotropic evolution of DC term (Reversible) Aging Response



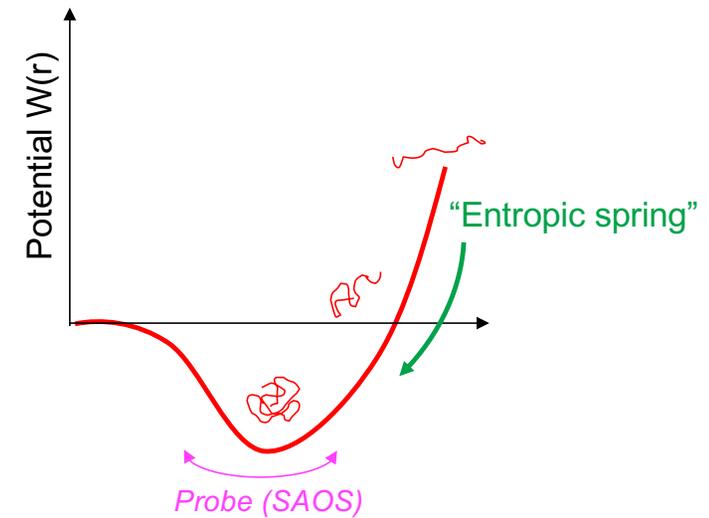
## PPR: Pump/Probe Rheometry (Schematic Thermodynamic Representation)



- **Drive** the system out of (local/instantaneous) equilibrium state using shear (**pump**) and **probe** resulting state *and its evolution in time* using small amplitude oscillatory shear (SAOS)
- In each case system seeks to minimize its Helmholtz Free Energy, *but different terms dominate*
- *Parallel Superposition Rheometry* (with Gabor transform for optimal separation of elastic, viscous and plastic contributions) in wet suspensions: Thixotropic Elastoviscoplastic (TEVP) fluids



Thixotropic Colloidal System:  $\Delta A = \Delta U - T\Delta S$



Polymer Solution/Melt:  $\Delta A = \Delta U - T\Delta S$